

Lecture 3 - Glycolysis and Gluconeogenesis

Chem 454: Regulatory Mechanisms in Biochemistry
University of Wisconsin-Eau Claire

1

Introduction

Glycolysis converts glucose ($C_6H_{12}O_6$) molecules to two molecules of pyruvic acid ($C_3H_4O_3$).

- Pyruvic acid is more oxidized than glucose
- The energy released from the oxidation is used to create 2 molecules of ATP from 2 ADP and 2 P_i
- This is an **anaerobic** process.
- Under anaerobic conditions the pyruvic acid can be **fermented** to lactic acid or to ethanol plus CO_2 .
- Under **aerobic** conditions, glucose is oxidized all the way to CO_2 and H_2O .

2

Introduction

Glucose can also be synthesized from molecules such as pyruvic acid or lactic acid.

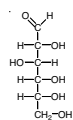
- This process is called **gluconeogenesis**.

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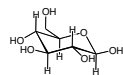
Introduction

Glucose is an important fuel for most organisms.

- In mammals, glucose is the preferred fuel source for the brain and the only fuel source for red blood cells.
- Almost all organisms use glucose



D-Glucose

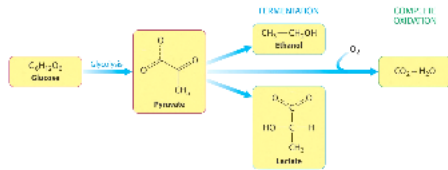


β -D-Glucose

4

Introduction

Fermentations provide usable energy in absence of oxygen.



5

Introduction

Obligate anaerobes

TABLE 16.2 Examples of pathogenic obligate anaerobes

Bacterium	Results of infection
<i>Clostridium tetani</i>	Tetanus (lockjaw)
<i>Clostridium botulinum</i>	Botulism (an especially severe type of food poisoning)
<i>Clostridium perfringens</i>	Gas gangrene (gas is produced as an end point of the fermentation, distorting and destroying the tissue)
<i>Bacteroides hensleyi</i>	Cat scratch fever (flu-like symptoms)
<i>Bacteroides fragilis</i>	Abdominal, pelvic, pulmonary, and blood infections

6

1. Glycolysis is Energy Conversion

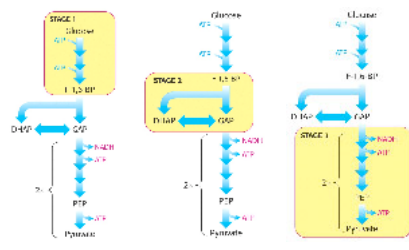
Glycolysis is an energy-conversion pathway in many organisms.

- ☛ The glycolytic pathway is common to virtually all organisms
- ☛ Both eukaryotes and prokaryotes
- ☛ In eukaryotes, it occurs in the cytosol

7

1. Glycolysis is Energy Conversion

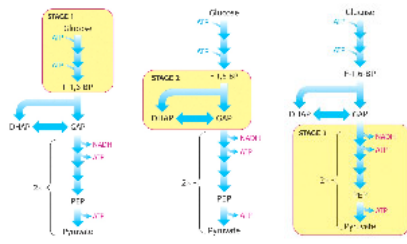
The glycolytic pathway is considered in three stages:



8

1. Glycolysis is Energy Conversion

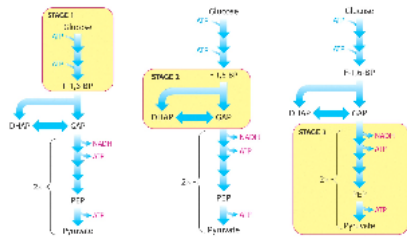
The glycolytic pathway is considered in three stages:



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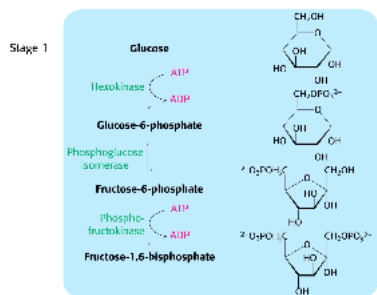
1. Glycolysis is Energy Conversion

The glycolytic pathway is considered in three stages:



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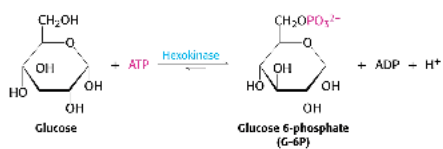
1. Stage 1



11

1.1. Hexokinase

Hexokinase traps glucose in the cell and begins glycolysis.

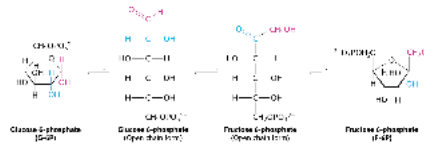


12

1.2 Phosphoglucose Isomerase

The formation of fructose 1,6-bisphosphate from glucose 6-phosphate

Phosphoglucose isomerase



13

1.2 Phosphofructokinase

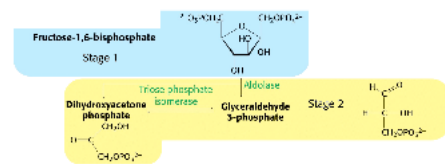
The formation of fructose 1,6-bisphosphate from glucose 6-phosphate

Phosphofructose kinase



14

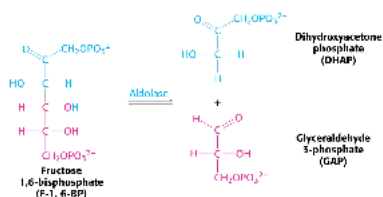
1. Stage 2



15

1.3. Aldolase

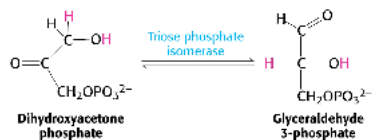
The six-carbon sugar is cleaved into two three-carbon fragments by aldolase.



16

1.4. Triose Phosphate Isomerase

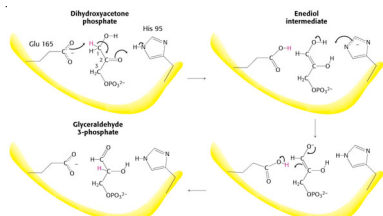
Triose phosphate isomerase salvages a three-carbon fragments



17

1.4. Triose Phosphate Isomerase

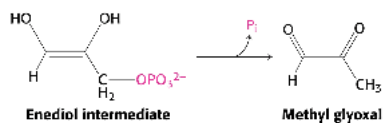
Triose phosphate isomerase salvages a three-carbon fragments



18

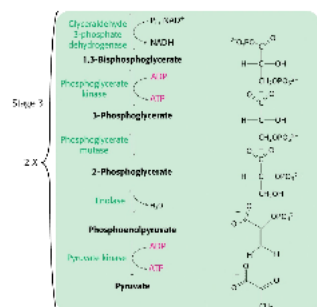
1.4. Triose Phosphate Isomerase

Triose phosphate isomerase is an example of a kinetically perfect enzyme.



19

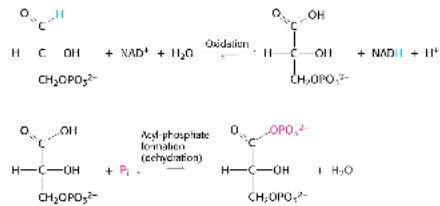
1. Stage 3



20

1.5. Glyceraldehyde 3-Phosphate Dehydrogenase

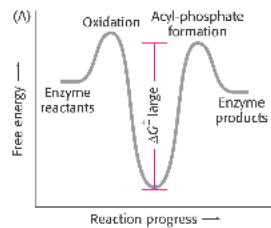
Energy transformation: Phosphorylation is coupled to the oxidation of glyceraldehyde 3-phosphate.



21

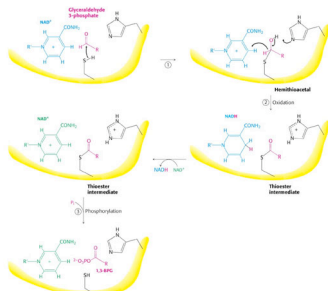
1.5. Glyceraldehyde 3-Phosphate Dehydrogenase

Energy transformation: Phosphorylation is coupled to the oxidation of glyceraldehyde 3-phosphate.



22

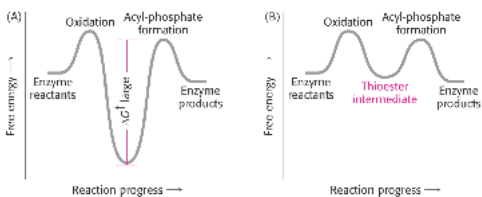
1.5. Glyceraldehyde 3-Phosphate Dehydrogenase



23

1.5. Glyceraldehyde 3-Phosphate Dehydrogenase

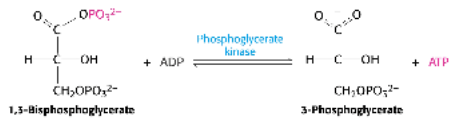
The enzyme-bound thioester intermediate reduces the activation energy for the second reaction:



24

1.6. Phosphoglycerate Kinase

The acyl phosphate in 1,3-bisphosphoglycerate has a high enough phosphoryl transfer potential to phosphorylate ADP to produce ATP:

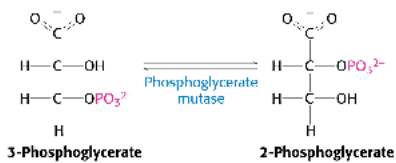


25

1.7. Phosphoglycerate Mutase

The next two reactions convert the remaining phosphate ester into a phosphate having a high phosphoryl transfer potential

• The first is an isomerization reaction

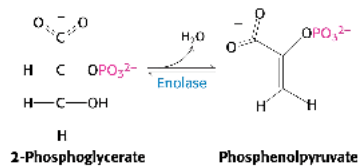


26

1.7. Enolase

The next two reactions convert the remaining phosphate ester into a phosphate having a high phosphoryl transfer potential

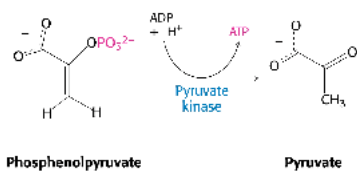
• The second is a dehydration (lyase) reaction



27

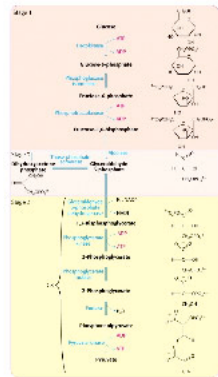
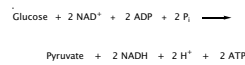
1.7. Pyruvate Kinase

The final reaction in the glycolytic pathway transfers the phosphate from phosphoenolpyruvate to ADP to produce ATP:



28

1.8 The Net Reaction



29

1.8. The Net Reaction

Summary of the reactions in the glycolytic pathway:

Step	Reaction	Enzyme	Electron type	ATP or NADH produced	ATP or NADH consumed
1	Glucose + 2 NAD ⁺ + 2 ADP + 2 P _i → Glucose-6-phosphate + 2 NADH + 2 H ⁺ + 2 ATP	Hexokinase	Electron transfer	2 NADH	2 NAD ⁺
2	Glucose-6-phosphate + 2 NAD ⁺ + 2 ADP + 2 P _i → Fructose-1,6-bisphosphate + 2 NADH + 2 H ⁺ + 2 ATP	Phosphofructokinase	Electron transfer	2 NADH	2 NAD ⁺
3	Fructose-1,6-bisphosphate → Glyceraldehyde-3-phosphate + Dihydroxyacetone phosphate	Aldolase	Aldehyde transfer	2 NADH	2 NAD ⁺
4	Glyceraldehyde-3-phosphate + 2 NAD ⁺ + 2 ADP + 2 P _i → 1,3-Bisphosphoglycerate + 2 NADH + 2 H ⁺ + 2 ATP	Glyceraldehyde-3-phosphate dehydrogenase	Electron transfer	2 NADH	2 NAD ⁺
5	1,3-Bisphosphoglycerate + 2 NAD ⁺ + 2 ADP + 2 P _i → 3-Phosphoglycerate + 2 NADH + 2 H ⁺ + 2 ATP	Phosphoglycerate kinase	Electron transfer	2 NADH	2 NAD ⁺
6	3-Phosphoglycerate + 2 NAD ⁺ + 2 ADP + 2 P _i → Pyruvate + 2 NADH + 2 H ⁺ + 2 ATP	Phosphoglycerate mutase, Enolase, Pyruvate kinase	Electron transfer	2 NADH	2 NAD ⁺

30

1.9. Maintaining Redox Balance

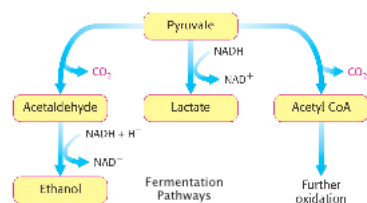
There is a problem:

- There are only catalytic quantities of NAD⁺ in the cell.
- In order to continue to use glycolysis to generate ATP, there needs to be some means of reoxidizing the NADH + H⁺ that is produced in glycolysis

31

1.9. Maintaining Redox Balance

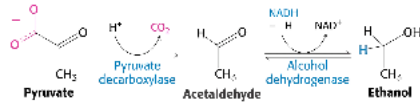
The solution to this problem lies in what happens to the pyruvate that is produced in glycolysis:



32

1.9.Maintaining Redox Balance

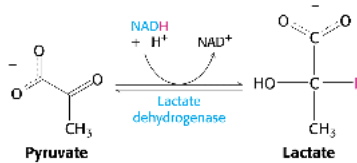
Ethanol fermentation is use by yeast and produces ethanol and CO₂.



33

1.9.Maintaining Redox Balance

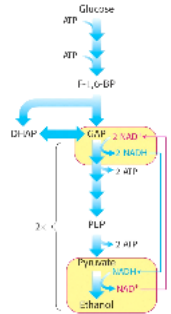
Lactic acid fermentation is use by bacteria and human muscles and produces lactate.



34

1.9.Maintaining Redox Balance

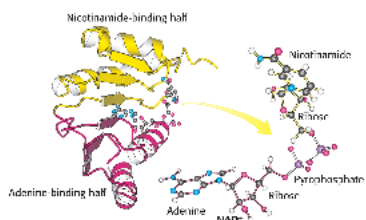
The fermentation pathways restore the redox balance:



35

1.10 NAD⁺ Binding

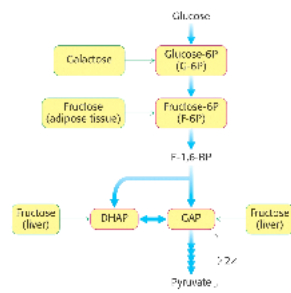
All three of the dehydrogenase in glycolysis and the fermentation pathways share a common domain for binding NAD⁺.



36

1.11 Other Points of Entry

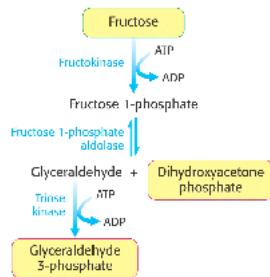
The entry of fructose and galactose into glycolysis.



37

1.11 Other Points of Entry

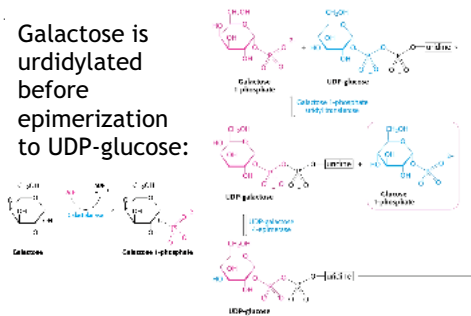
Fructose is phosphorylated to fructose 1-phosphate in the liver.



38

1.11 Other Points of Entry

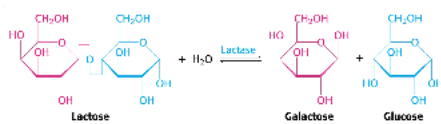
Galactose is uridylylated before epimerization to UDP-glucose:



39

1.12 Lactose Intolerance

Many adults are intolerant of milk because they are deficient in lactase



40

1.13 Galactose is Highly Toxic

📍 Disruption of galactose metabolism is called *galactosemia*.

📍 Usually due to loss of uridyl transferase activity

📍 Symptoms include

- 📍 Failure to thrive infants
- 📍 Enlarged liver and jaundice, sometimes cirrhosis
- 📍 Cataracts
- 📍 Mental retardation

41

41

2. Control of Glycolysis

Two major needs of the the cell influence the flow of material from glucose to pyruvate:

- 📍 The need for ATP (energy)
- 📍 The need for building blocks for biosynthesis

42

42

2. Control of Glycolysis

In metabolic pathways, control is focused on those steps in the pathway that are irreversible.

TABLE 22-2 *Enzymes of Glycolysis*

Step	Enzyme	Reaction	Reaction Type	ΔG° (kJ/mol)	ΔG (kJ/mol)
1	Hexokinase (EC 2.7.1.1)	Glucose + ATP → Glucose-6-phosphate + ADP	Phosphorylation	-16.7	-16.7
2	Glucose-6-phosphate dehydrogenase (EC 1.1.1.42)	Glucose-6-phosphate + H ₂ O → 6-Phosphogluconate + NADPH + H ⁺	Oxidation	-16.7	-16.7
3	Phosphoglucomutase (EC 5.4.2.1)	Glucose-6-phosphate ↔ Fructose-6-phosphate	Isomerization	0	0
4	Phosphofructokinase (EC 2.7.1.11)	Fructose-6-phosphate + ATP → Fructose-1,6-bisphosphate + ADP	Phosphorylation	-14.2	-14.2
5	Aldehyde dehydrogenase (EC 1.2.1.12)	Fructose-1,6-bisphosphate + H ₂ O → Dihydroxyacetone phosphate + NADPH + H ⁺	Oxidation	-14.2	-14.2
6	Triose phosphate isomerase (EC 5.3.1.1)	Dihydroxyacetone phosphate ↔ Glyceraldehyde-3-phosphate	Isomerization	0	0
7	Glyceraldehyde-3-phosphate dehydrogenase (EC 1.2.1.12)	Glyceraldehyde-3-phosphate + NAD ⁺ → 1,3-Bisphosphoglycerate + NADH + H ⁺	Oxidation	-14.2	-14.2
8	Phosphoglycerate kinase (EC 2.7.2.3)	1,3-Bisphosphoglycerate + ADP → 3-Phosphoglycerate + ATP	Phosphorylation	-14.2	-14.2
9	Phosphoglycerate mutase (EC 5.4.2.1)	3-Phosphoglycerate ↔ 2-Phosphoglycerate	Isomerization	0	0
10	Enolase (EC 4.2.1.11)	2-Phosphoglycerate → Phosphoenolpyruvate + H ₂ O	Dehydration	-14.2	-14.2

43

43

2. Control of Glycolysis

The different levels of control have different response times:

Level of Control	Response Time
Allosteric	milleseconds
Phosphorylation	seconds
Transcriptional	hours

44

44

2.1 Phosphofructokinase

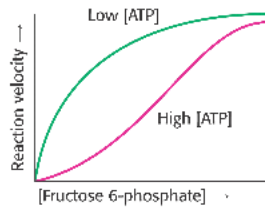
Phosphofructokinase is the key enzyme in the control of glycolysis.



45

2.1 Phosphofructokinase

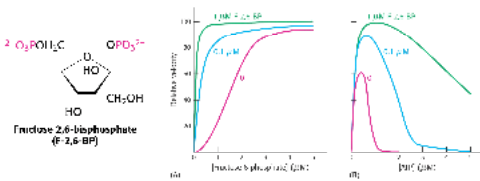
Allosteric regulation of phosphofructokinase by ATP



46

2.1 Phosphofructokinase

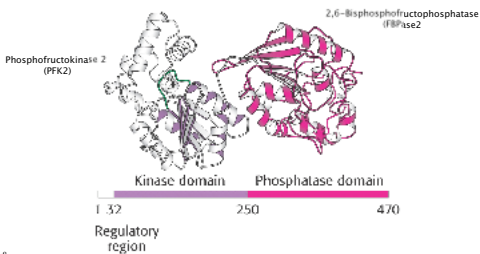
Phosphofructokinase is also regulated by fructose 2,6-bisphosphate:



47

2.2. Fructose 2,6-bisphosphate

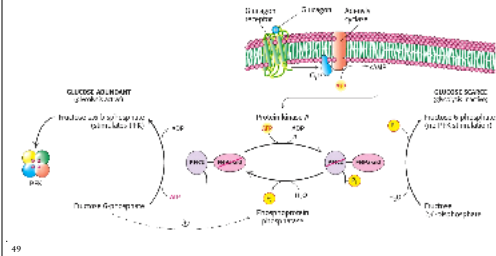
A regulated bifunctional enzyme synthesizes and degrades fructose 2,6-bisphosphate:



48

2.2. Fructose 2,6-bisphosphate

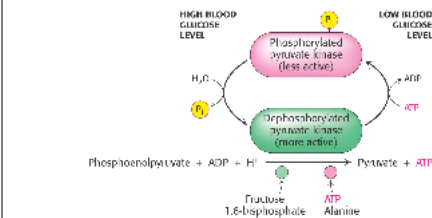
A regulated bifunctional enzyme synthesizes and degrades fructose 2,6-bisphosphate:



49

2.3 Hexokinase and Pyruvate Kinase

Hexokinase and pyruvate kinase also set the pace of glycolysis.



50

2.4. Glucose Transporters

A family of transporters enables glucose to enter and leave animal cells.

Glucose transporter	Tissue location	K_m	Comments
GLUT-1	All mammalian tissues	1-10 mM	Basal glucose uptake
GLUT-2	Liver and pancreatic β cells	15-40 mM	In the pancreas, plays a role in regulation of β cells. In the liver, releases glucose from the blood
GLUT-3	All mammalian tissues	1-10 mM	Basal glucose uptake
GLUT-4	Muscle and fat cells	1-10 mM	Amounts to increase plasma membrane, increases rate of glucose uptake
GLUT-5	Small intestine		Primarily a glucose transporter

51

3. Gluconeogenesis

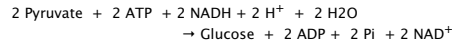
Glucose can be synthesized from noncarbohydrate precursors.

- ☞ The brain has a strong preference for glucose, while the red blood cells have an absolute requirement for glucose.
- ☞ The brain needs 120 g of glucose/day
- ☞ The liver has about a 190 g store of glucose as glycogen. (About a 1 day's supply)
- ☞ Glucose can be synthesized in the liver from pyruvate, glycerol and amino acids.

52

3.1. Gluconeogenesis

Gluconeogenesis is not the reverse of glycolysis.



The ΔG° for this reaction is +20 kcal/mol

53

3.1. Gluconeogenesis

The three kinase reactions are the ones with the greatest positive free energies in the reverse directions

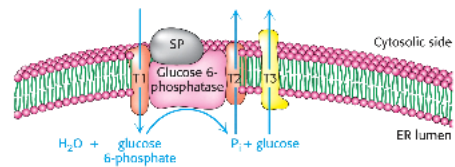
TABLE 3-1 Reactions of Gluconeogenesis

Step	Reaction	Enzyme	Reaction type	ΔG° (kcal/mol)	ΔG° (kJ/mol)
1	$\text{Pyruvate} + \text{H}_2\text{O} \rightarrow \text{Oxaloacetate} + \text{H}^+$	Pyruvate carboxylase	Carboxylation	-33.0	-138
2	$\text{Oxaloacetate} + \text{PEP} \rightarrow \text{Phosphoenolpyruvate} + \text{H}_2\text{O}$	Phosphoenolpyruvate carboxykinase	Decarboxylation	-1.3	-5.4
3	$\text{Phosphoenolpyruvate} + \text{H}_2\text{O} \rightarrow \text{2-Phosphoglycerate}$	Phosphoenolpyruvate hydratase	Hydration	-1.3	-5.4
4	$\text{2-Phosphoglycerate} \rightarrow \text{3-Phosphoglycerate}$	Phosphoglycerate kinase	Phosphorylation	-0.4	-1.7
5	$\text{3-Phosphoglycerate} \rightarrow \text{1,3-Bisphosphoglycerate}$	Phosphoglycerate kinase	Phosphorylation	-0.4	-1.7
6	$\text{1,3-Bisphosphoglycerate} + \text{H}_2\text{O} \rightarrow \text{3-Carboxyphosphoglycerate}$	Phosphoglycerate kinase	Hydration	-0.4	-1.7
7	$\text{3-Carboxyphosphoglycerate} \rightarrow \text{2-Carboxyphosphoglycerate}$	Phosphoglycerate kinase	Phosphorylation	-0.4	-1.7
8	$\text{2-Carboxyphosphoglycerate} \rightarrow \text{Phosphoenolpyruvate}$	Phosphoenolpyruvate carboxykinase	Decarboxylation	-1.3	-5.4
9	$\text{Phosphoenolpyruvate} + \text{H}_2\text{O} \rightarrow \text{2-Phosphoglycerate}$	Phosphoenolpyruvate hydratase	Hydration	-1.3	-5.4
10	$\text{2-Phosphoglycerate} \rightarrow \text{3-Phosphoglycerate}$	Phosphoglycerate kinase	Phosphorylation	-0.4	-1.7
11	$\text{3-Phosphoglycerate} \rightarrow \text{1,3-Bisphosphoglycerate}$	Phosphoglycerate kinase	Phosphorylation	-0.4	-1.7
12	$\text{1,3-Bisphosphoglycerate} + \text{H}_2\text{O} \rightarrow \text{3-Carboxyphosphoglycerate}$	Phosphoglycerate kinase	Hydration	-0.4	-1.7
13	$\text{3-Carboxyphosphoglycerate} \rightarrow \text{2-Carboxyphosphoglycerate}$	Phosphoglycerate kinase	Phosphorylation	-0.4	-1.7
14	$\text{2-Carboxyphosphoglycerate} \rightarrow \text{Phosphoenolpyruvate}$	Phosphoenolpyruvate carboxykinase	Decarboxylation	-1.3	-5.4
15	$\text{Phosphoenolpyruvate} + \text{H}_2\text{O} \rightarrow \text{2-Phosphoglycerate}$	Phosphoenolpyruvate hydratase	Hydration	-1.3	-5.4
16	$\text{2-Phosphoglycerate} \rightarrow \text{3-Phosphoglycerate}$	Phosphoglycerate kinase	Phosphorylation	-0.4	-1.7
17	$\text{3-Phosphoglycerate} \rightarrow \text{1,3-Bisphosphoglycerate}$	Phosphoglycerate kinase	Phosphorylation	-0.4	-1.7
18	$\text{1,3-Bisphosphoglycerate} + \text{H}_2\text{O} \rightarrow \text{3-Carboxyphosphoglycerate}$	Phosphoglycerate kinase	Hydration	-0.4	-1.7
19	$\text{3-Carboxyphosphoglycerate} \rightarrow \text{2-Carboxyphosphoglycerate}$	Phosphoglycerate kinase	Phosphorylation	-0.4	-1.7
20	$\text{2-Carboxyphosphoglycerate} \rightarrow \text{Phosphoenolpyruvate}$	Phosphoenolpyruvate carboxykinase	Decarboxylation	-1.3	-5.4

54

3.1. Gluconeogenesis

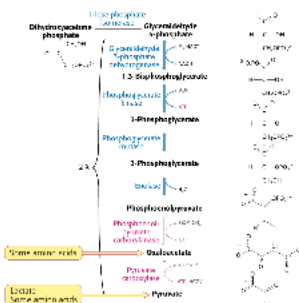
The hexokinase and phosphofructokinase reactions can be reversed simply with a phosphatase



55

3.2. Formation of Phosphoenolpyruvate

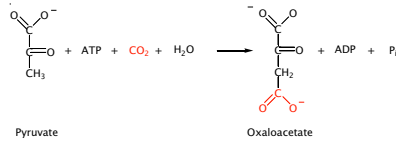
Reversing the pyruvate kinase reaction is not as easily done



56

3.2. Formation of Phosphoenolpyruvate

The conversion of pyruvate into phosphoenolpyruvate begins with the formation of oxaloacetate.

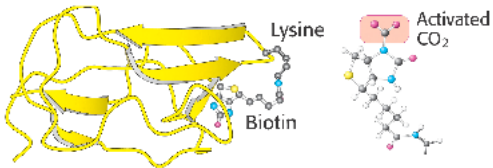


57

57

3.2. Formation of Phosphoenolpyruvate

Pyruvate kinase uses the biotin cofactor to activate the CO₂

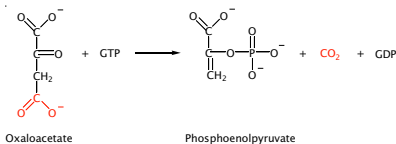


58

58

3.2. Formation of Phosphoenolpyruvate

The formation of phosphoenolpyruvate from oxaloacetate is driven both by the hydrolysis of GTP and a decarboxylation

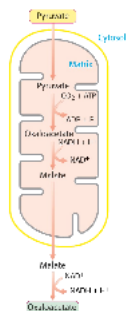


59

59

3.3. Oxaloacetate Shuttle

Oxaloacetate is synthesized in the mitochondria and is shuttled into the cytosol where it is converted into phosphoenolpyruvate



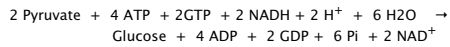
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3.6. "High-Energy" Phosphate Bonds

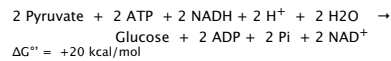
Six high-energy phosphate bonds are spent in synthesizing glucose from pyruvate.

Gluconeogenesis:



$$\Delta G^{\circ} = -9 \text{ kcal/mol}$$

Reverse of Glycolysis:



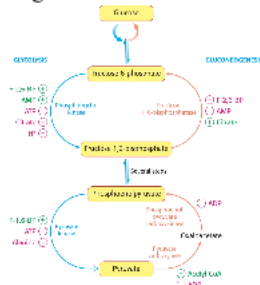
$$\Delta G^{\circ} = +20 \text{ kcal/mol}$$

61

61

4. Regulation of Glycolysis and Gluconeogenesis

Reciprocal regulation of glycolysis and gluconeogenesis in the liver

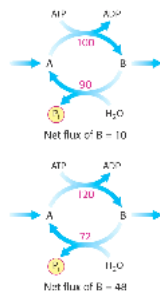


62

62

4.1. Substrate Cycles

Substrate cycles amplify metabolic signals and produce heat.

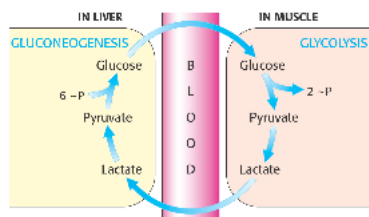


63

63

4.2. Lactate and Alanine

Lactate and alanine formed by contracting muscle are used by other organs

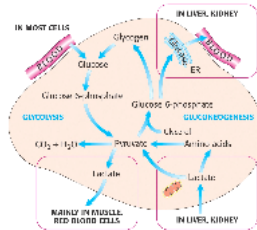


64

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4.2. Lactate and Alanine

Lactate and alanine formed by contracting muscle are used by other organs



65

4.3. Evolution of Glycolysis and Gluconeogenesis

💡 Glycolysis and Gluconeogenesis are evolutionarily intertwined.

66