Topic 10 Drugs of the Nervous System

Ch 19,20 Patrick Part VI- Nervous system -Corey

Contents

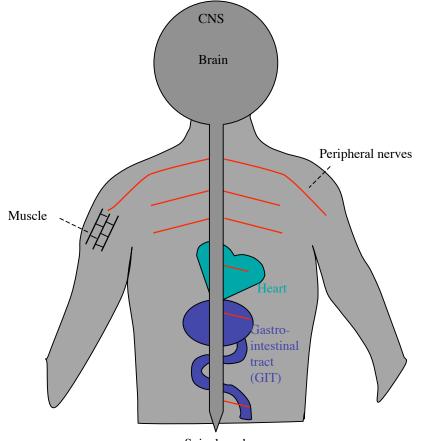
Part 1: Cholinergics & anticholinesterases

- 1. Nerve Transmission
- 2. Neurotransmitter
- 3. Transmission
- 4. Cholinergic receptors
 - 4.1. Nicotinic receptor
 - 4.2. Muscarinic receptor G Protein coupled receptor
- 5. Cholinergic agonists
 - 5.1. Acetylcholine as an agonist
 - 5.2. Nicotine and muscarine as cholinergic agonists
 - 5.3. Requirements for cholinergic agonists
- 6. SAR for acetylcholine
- 7. Binding site (muscarinic)
- 8. Active conformation of acetylcholine
- 9. Instability of acetylcholine
- 10. Design of cholinergic agonists
- 11. Uses of cholinergic agonists

CHOLINERGIC NERVOUS SYSTEM

1. Nerve Transmission

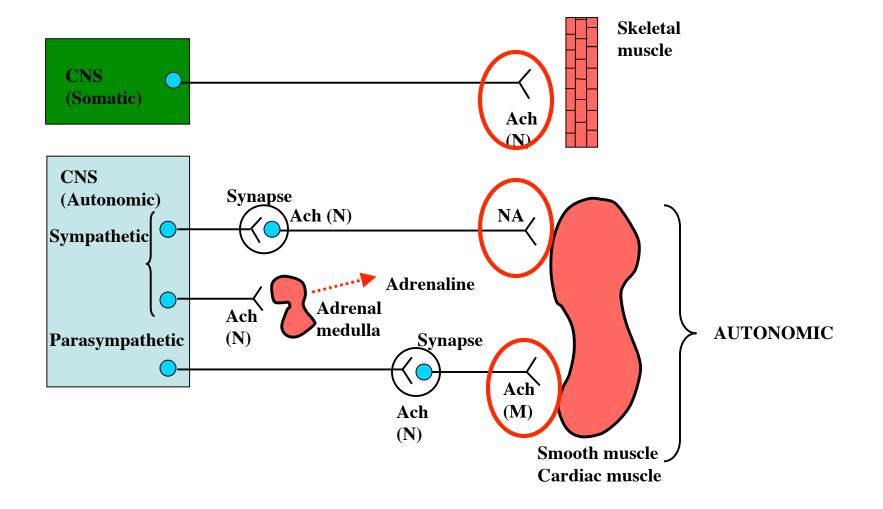
Peripheral nervous system



Spinal cord

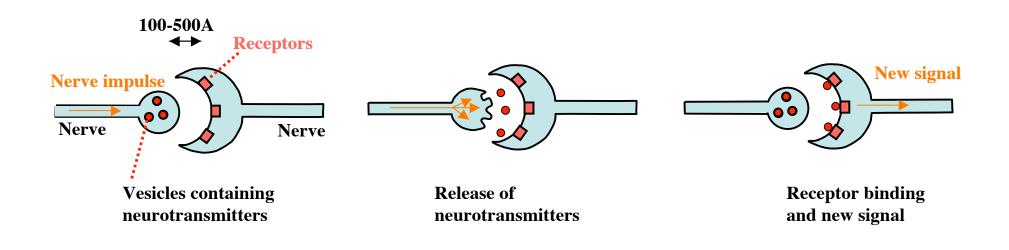
1. Nerve Transmission

Peripheral nervous system



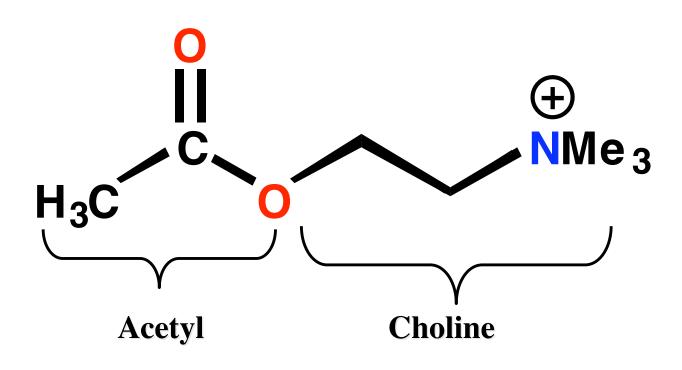
1. Nerve Transmission

Synapses

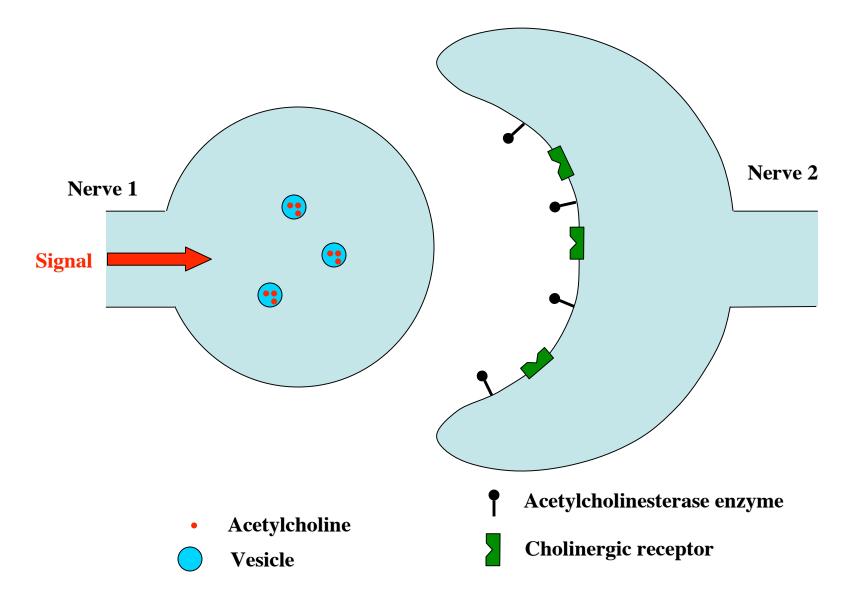




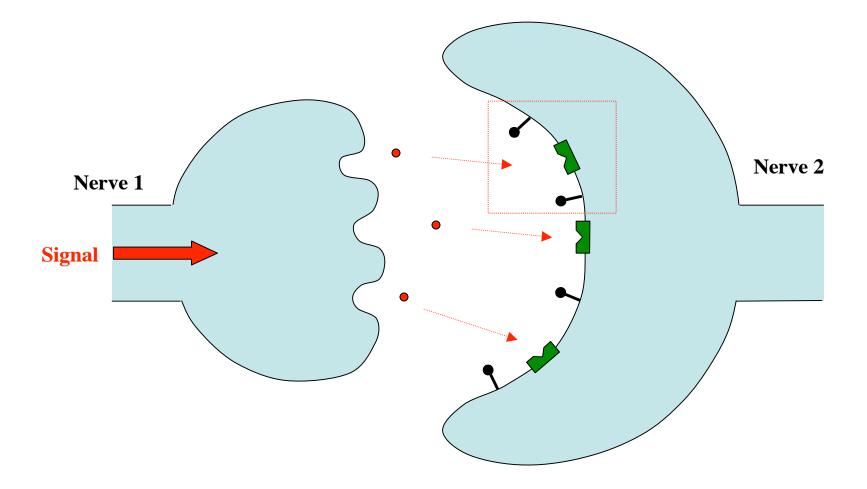
Acetylcholine (Ach)

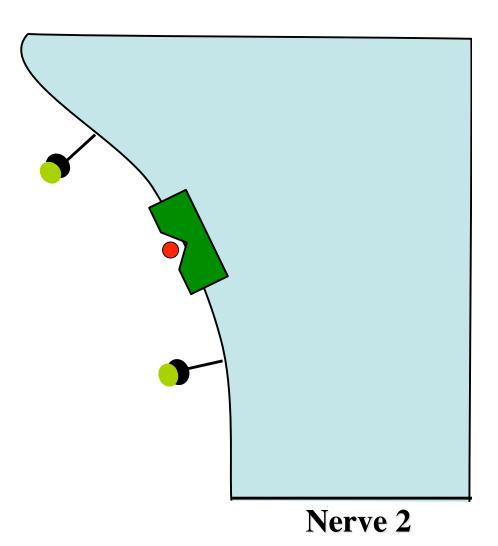


Signal in nerve 1

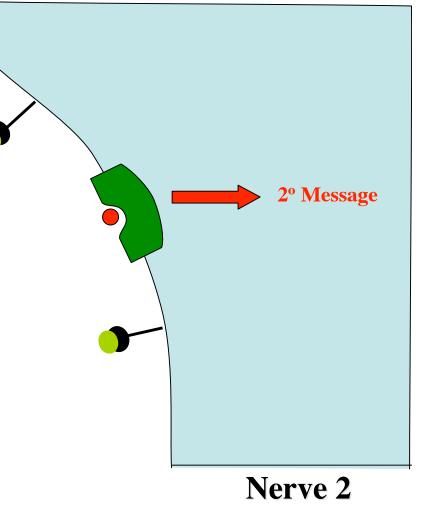


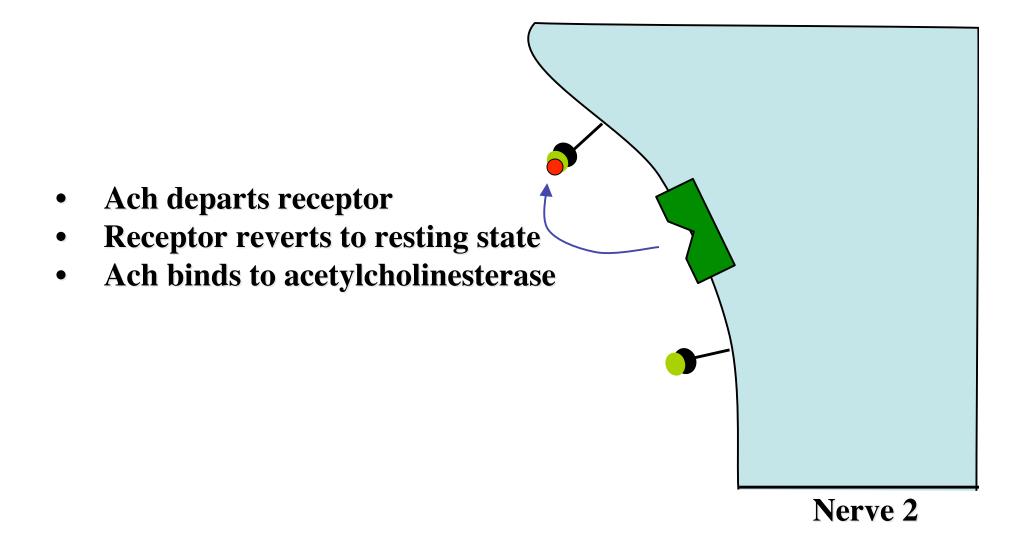
Vesicles fuse with membrane and release Ach



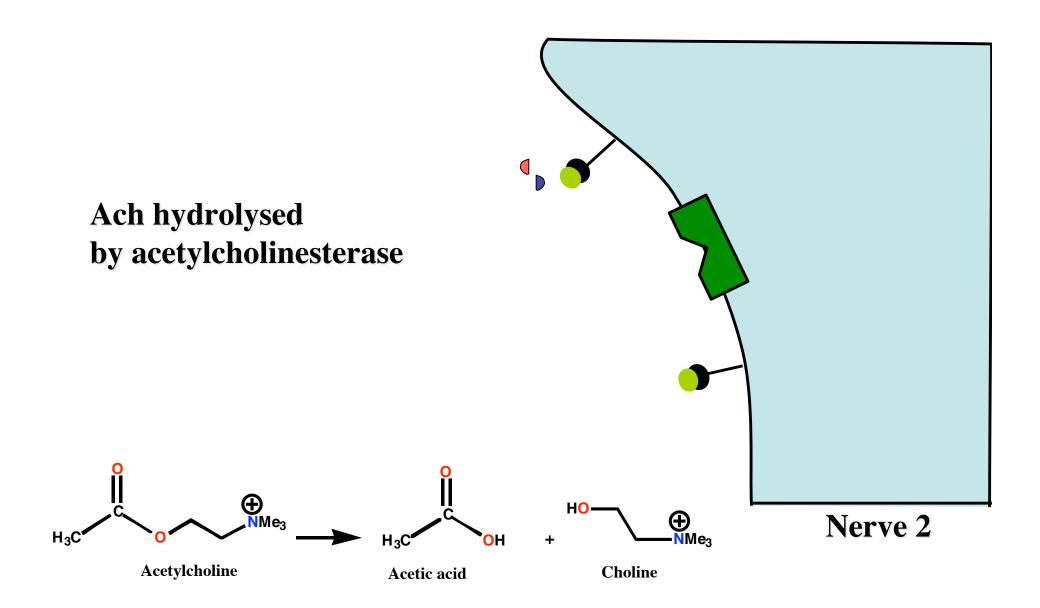


- Receptor binds Ach
- Induced fit triggers 2° message
- Triggers firing of nerve 2
- Ach undergoes no reaction

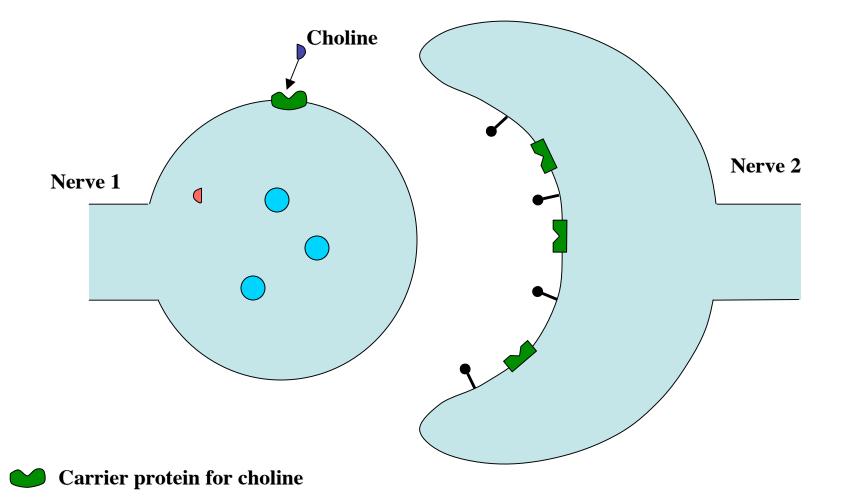




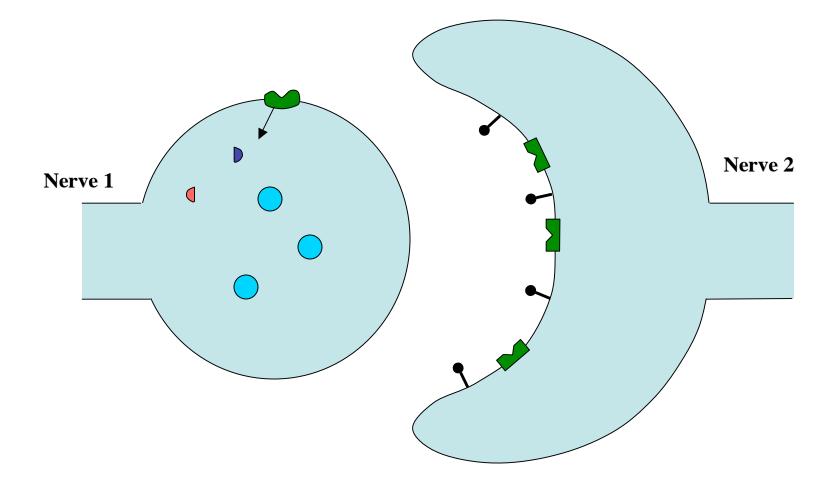




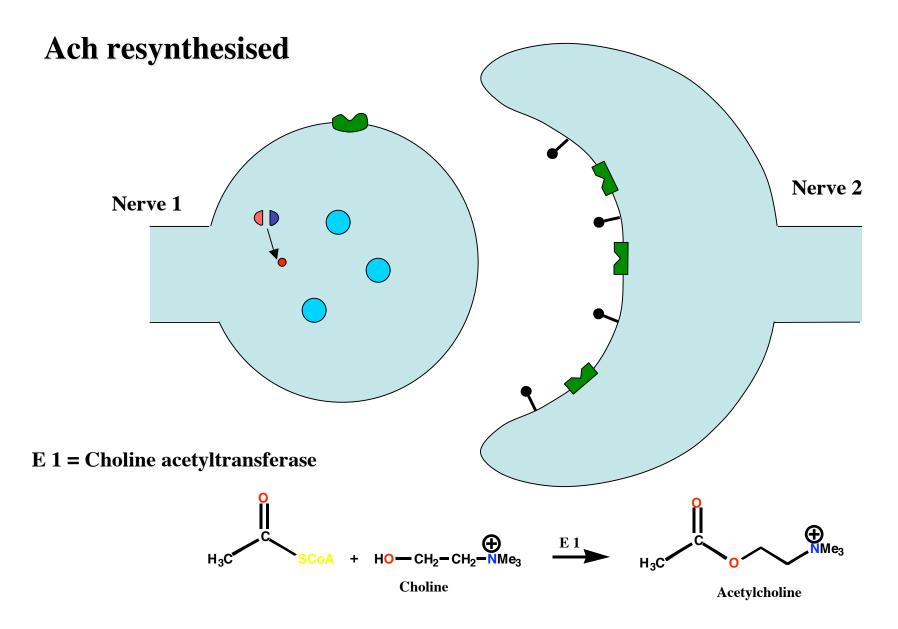
Choline binds to carrier protein



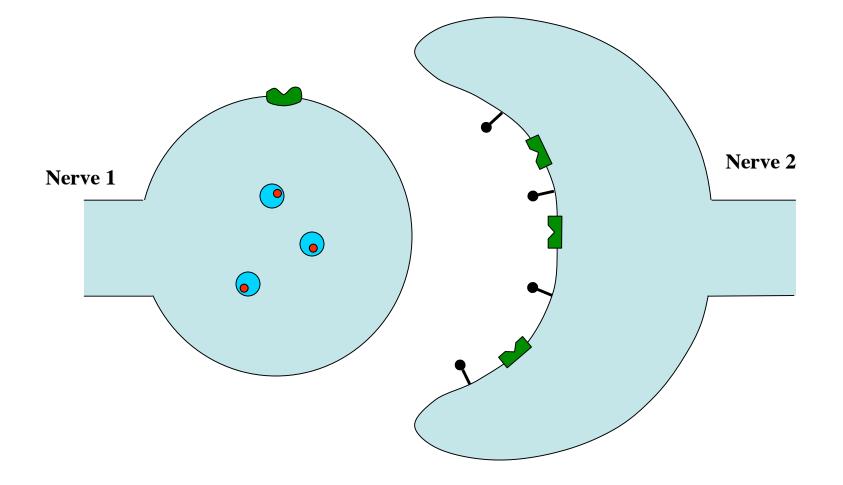
Choline transported into nerve







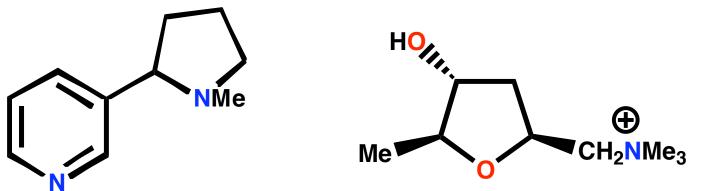
Ach repackaged in vesicles



4. Cholinergic receptors

Receptor types

- Not all cholinergic receptors are identical
- Two types of cholinergic receptor nicotinic and muscarinic
- Named after natural products showing receptor selectivity



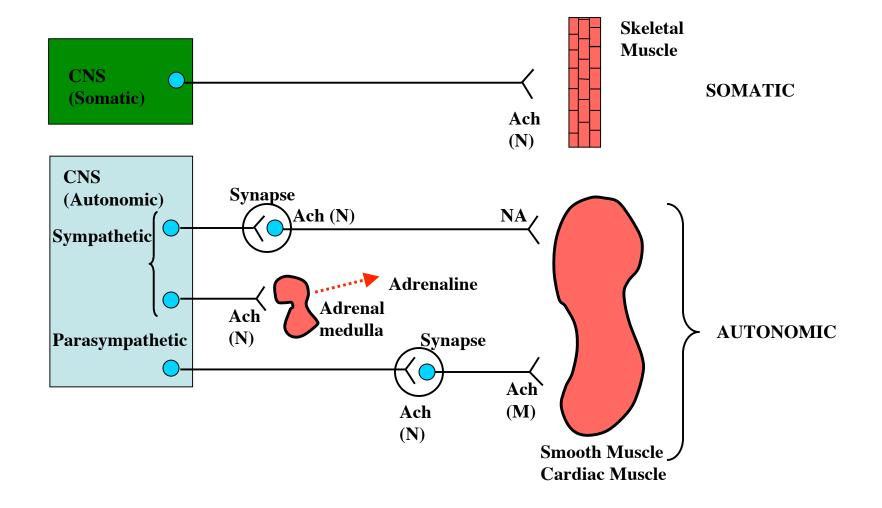
Nicotine

L-(+)-Muscarine

Activates cholinergic receptors at nerve synapses and on skeletal muscle Activates cholinergic receptors on smooth muscle and cardiac muscle

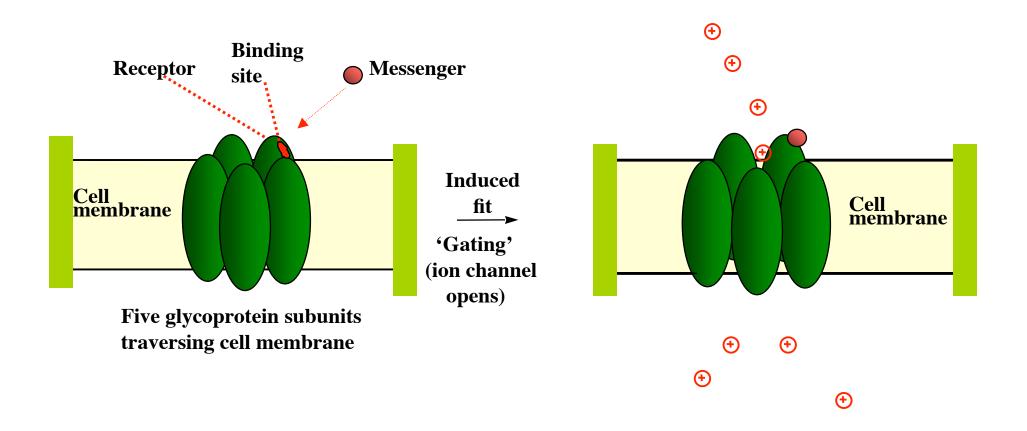
Acetylcholine is natural messenger for both receptor types

Peripheral nervous system



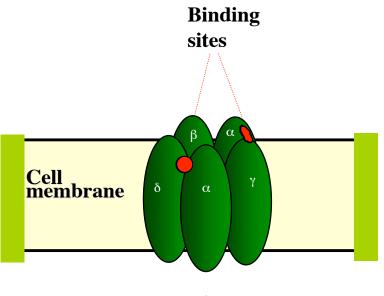
4.1 Nicotinic receptor

Control of Cationic Ion Channel:

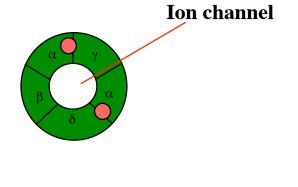




The binding sites



 $2x\alpha, \beta, \gamma, \delta$ subunits

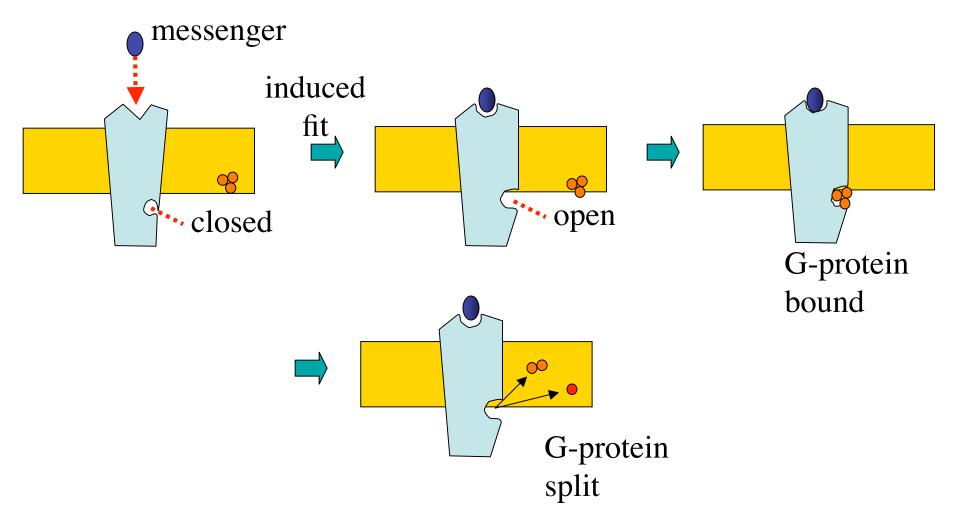


• Two ligand binding sites mainly on α -subunits

4.2 Muscarinic receptor - G Protein coupled receptor

Activation of a signal protein

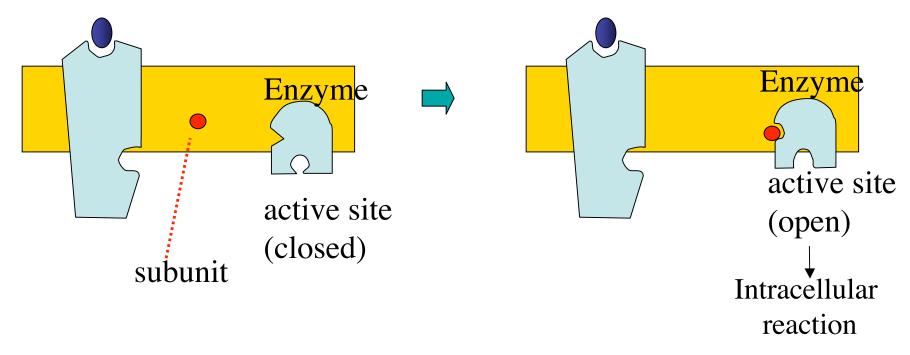
- Receptor binds messenger leading to an induced fit
- Opens a binding site for a signal protein (G-protein)



4.2 Muscarinic receptor - G Protein coupled receptor

Activation of membrane bound enzyme

- G-Protein is split and subunit activates a membrane bound enzyme
- Subunit binds to an allosteric binding site on enzyme
- Induced fit results in opening of an active site
- Intracellular reaction is catalysed

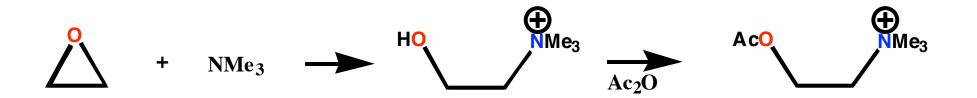


5. Cholinergic agonists

5.1 Acetylcholine as an agonist

Advantages

- Natural messenger
- Easily synthesized



Disadvantages

- Easily hydrolysed in stomach (acid catalysed hydrolysis)
- Easily hydrolysed in blood (esterases)
- No selectivity between receptor types
- No selectivity between different target organs

5. Cholinergic agonists

5.2 Nicotine and muscarine as cholinergic agonists

Advantages

- More stable than Ach
- Selective for main cholinergic receptor types
- Selective for different organs

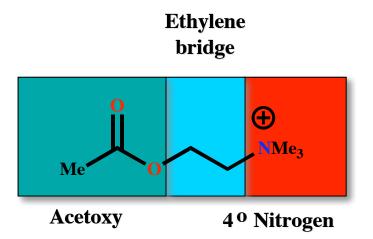
Disadvantages

- Activate receptors for other chemical messengers
- Side effects

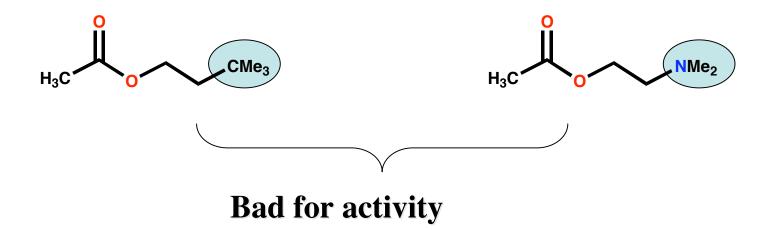
5. Cholinergic agonists

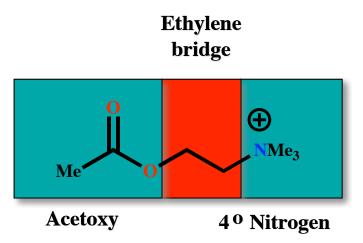
5.3 Requirements for cholinergic agonists

- Stability to stomach acids and esterases
- Selectivity for cholinergic receptors
- Selectivity between muscarinic and nicotinic receptors
- Knowledge of binding site
- SAR for acetylcholine

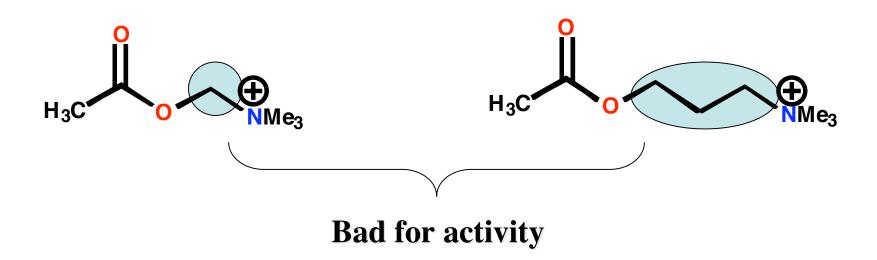


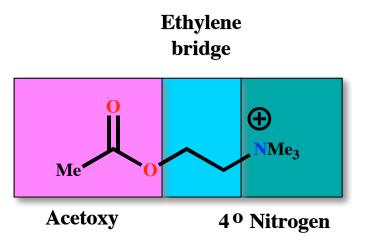
Quaternary nitrogen is essential



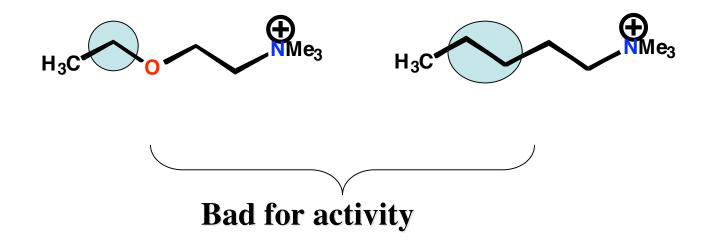


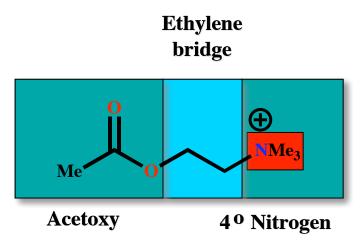
- Distance from quaternary nitrogen to ester is important
- Ethylene bridge must be retained



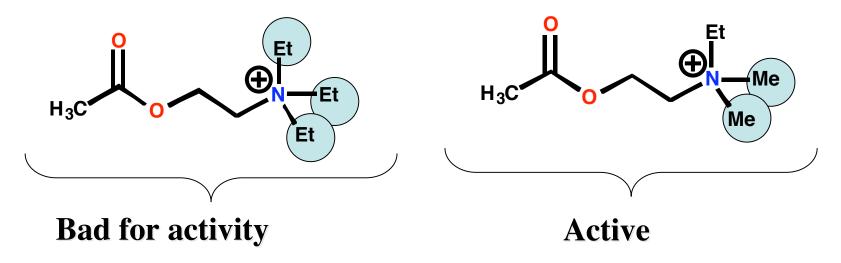


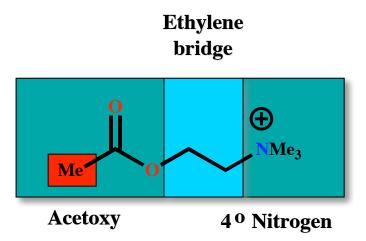
Ester is important



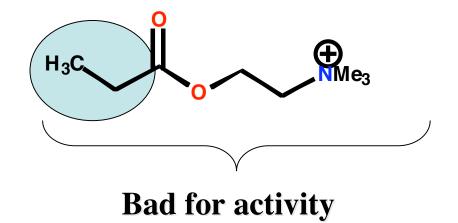


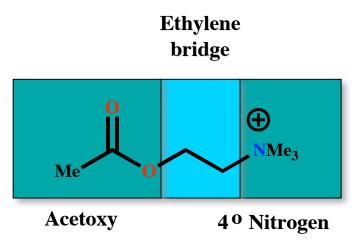
Minimum of two methyl groups on quaternary nitrogen





Methyl group of acetoxy group cannot be extended

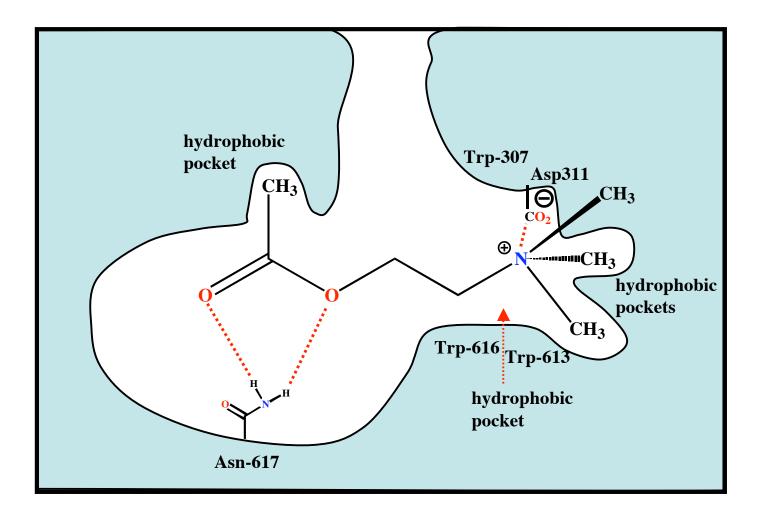




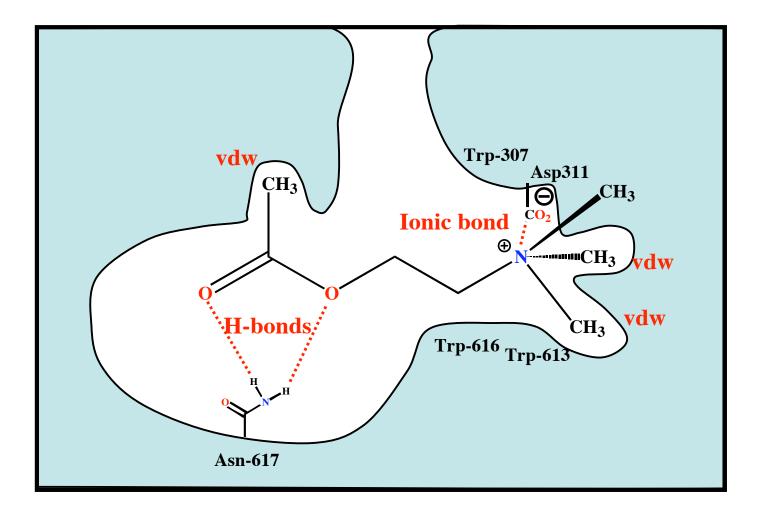
Conclusions:

- Tight fit between Ach and binding site
- Methyl groups fit into small hydrophobic pockets
- Ester interacting by H-bonding
- Quaternary nitrogen interacting by ionic bonding

7. Binding site (muscarinic)

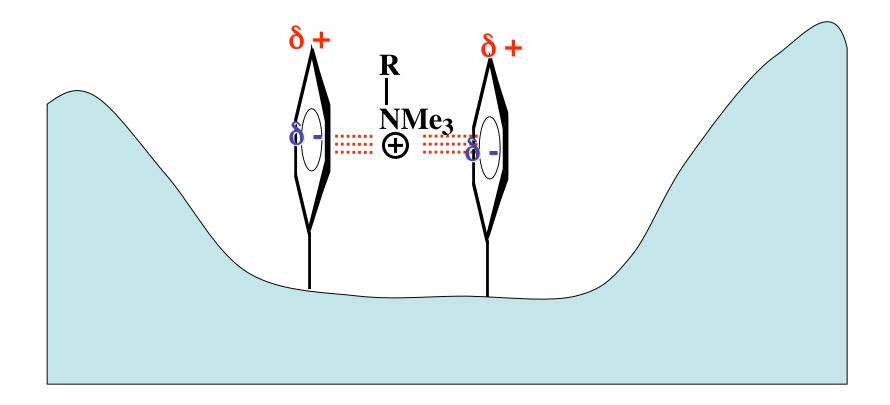


7. Binding site (muscarinic)

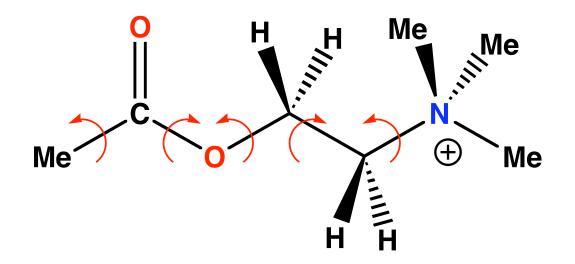


7. Binding site (muscarinic)

- Possible induced dipole dipole interaction between quaternary nitrogen and hydrophobic aromatic rings in binding site
- N⁺ induces dipole in aromatic rings



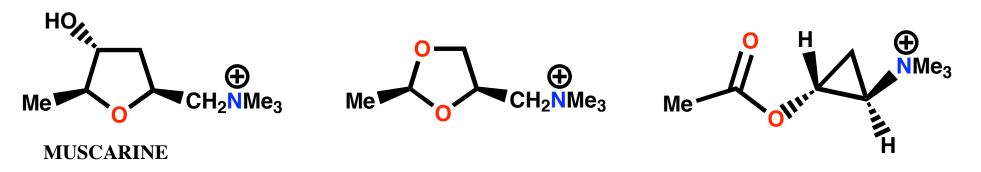
8. Active conformation of acetylcholine



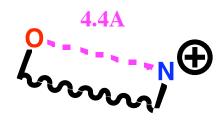
- Several freely rotatable single bonds
- Large number of possible conformations
- Active conformation does not necessarily equal the most stable conformation

8. Active conformation of acetylcholine

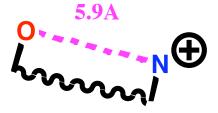
Rigid Analogues of acetylcholine



- Rotatable bonds 'locked' within ring
- Restricts number of possible conformations
- Defines separation of ester and N

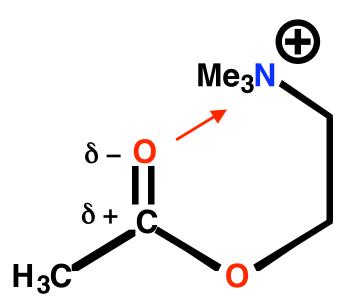


Muscarinic receptor



Nicotinic receptor

9. Instability of acetylcholine



- Neighbouring group participation
- Increases electrophilicity of carbonyl group
- Increases sensitivity to nucleophiles

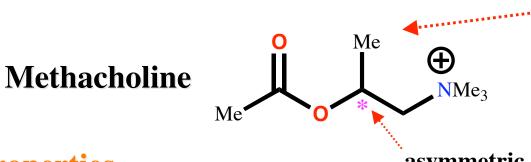
Requirements

- Correct size
- Correct pharmacophore ester and quaternary nitrogen
- Increased stability to acid and esterases
- Increased selectivity

Use of steric shields

Rationale

- Shields protect ester from nucleophiles and enzymes
- Shield size is important
- Must be large enough to hinder hydrolysis
- Must be small enough to fit binding site

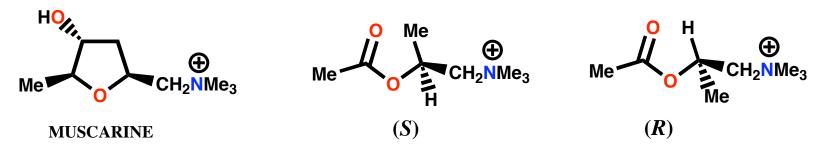


hinders binding to esterases and provides a shield to nucleophilic attack

Properties

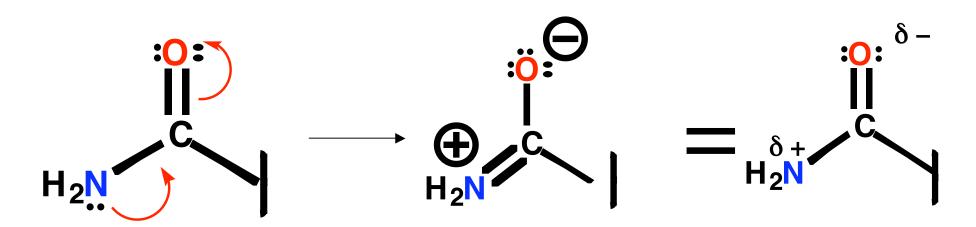
asymmetric centre

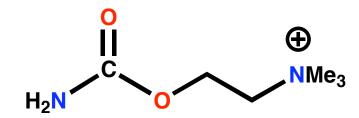
- Three times more stable than acetylcholine
- Increasing the shield size increases stability but decreases activity
- Selective for muscarinic receptors over nicotinic receptors
- S-enantiomer is more active than the R-enantiomer
- Stereochemistry matches muscarine
- Not used clinically



Use of electronic factors

- Replace ester with urethane
- Stabilises the carbonyl group



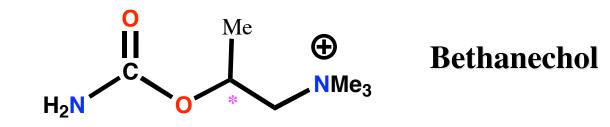


Carbachol

Properties

- Resistant to hydrolysis
- Long lasting
- NH₂ and CH₃ are equal sizes. Both fit the hydrophobic pocket
- NH₂ = bio-isostere
- Muscarinic activity = nicotinic activity
- Used topically for glaucoma

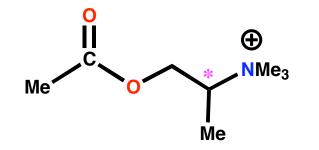
Steric + Electronic factors



Properties

- Very stable
- Orally active
- Selective for the muscarinic receptor
- Used to stimulate GI tract and urinary bladder after surgery

Nicotinic selective agonist



* asymmetric centre

Nicotinic selective agonists

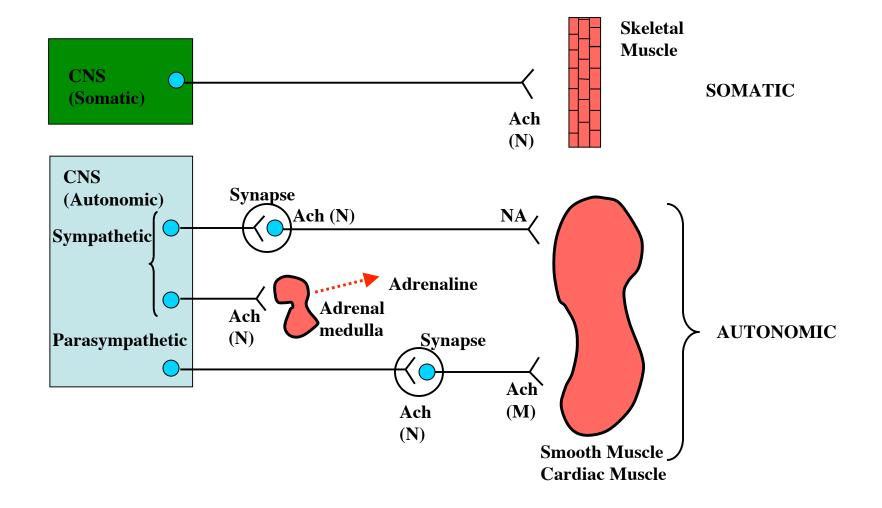
Treatment of myasthenia gravis

- lack of acetylcholine at skeletal muscle causing weakness

Muscarinic selective agonists

- Treatment of glaucoma
- Switching on GIT and urinary tract after surgery
- Treatment of certain heart defects. Decreases heart muscle activity and decreases heart rate

Peripheral nervous system



Contents

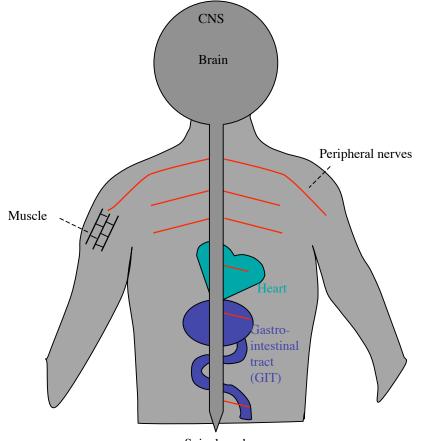
Part 1: Cholinergics & anticholinesterases

- 1. Nerve Transmission
- 2. Neurotransmitter
- 3. Transmission process
- 4. Cholinergic receptors
 - 4.1. Nicotinic receptor
 - 4.2. Muscarinic receptor G Protein coupled receptor
- 5. Cholinergic agonists
 - 5.1. Acetylcholine as an agonist
 - 5.2. Nicotine and muscarine as cholinergic agonists
 - 5.3. Requirements for cholinergic agonists
- 6. SAR for acetlcholine
- 7. Binding site (muscarinic)
- 8. Active conformation of acetylcholine
- 9. Instability of acetylcholine
- 10. Design of cholinergic agonists
- 11. Uses of cholinergic agonists

CHOLINERGIC NERVOUS SYSTEM

1. Nerve Transmission

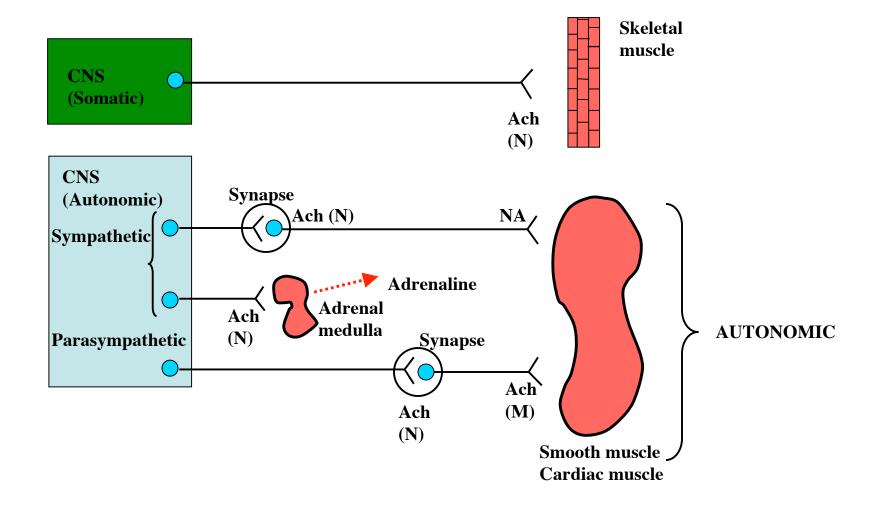
Peripheral nervous system



Spinal cord

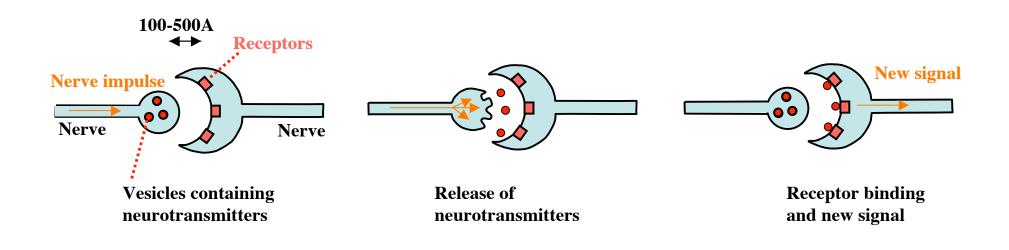
1. Nerve Transmission

Peripheral nervous system



1. Nerve Transmission

Synapses

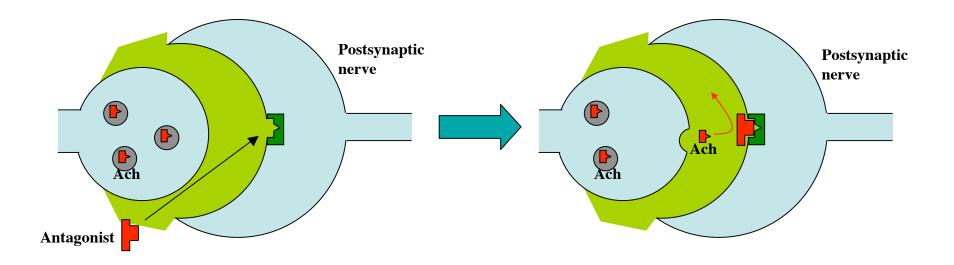


Contents

Part 2: Cholinergics & anticholinesterases

- 12. Cholinergic Antagonists (Muscarinic receptor)
 - 12.1. Atropine
 - 12.2. Hyoscine (scopolamine)
 - 12.3. Comparison of atropine with acetylcholine
 - 12.4. Analogues of atropine
 - 12.5. Simplified Analogues
 - 12.6. SAR for Antagonists
 - 12.7. Binding Site for Antagonists
- 13. Cholinergic Antagonists (Nicotinic receptor)
 - 13.1. Curare
 - 13.2. Binding
 - 13.3. Analogues of tubocurarine

- Drugs which bind to cholinergic receptor but do not activate it
- Prevent acetylcholine from binding
- Opposite clinical effect to agonists lower activity of acetylcholine



Clinical Effects

- Decrease of saliva and gastric secretions
- Relaxation of smooth muscle
- Decrease in motility of GIT and urinary tract
- Dilation of pupils

Uses

- Shutting down digestion for surgery
- Ophthalmic examinations
- Relief of peptic ulcers
- Treatment of Parkinson's Disease
- Anticholinesterase poisoning
- Motion sickness

12.1 Atropine

easily racemised

Racemic form of hyoscyamine

Me

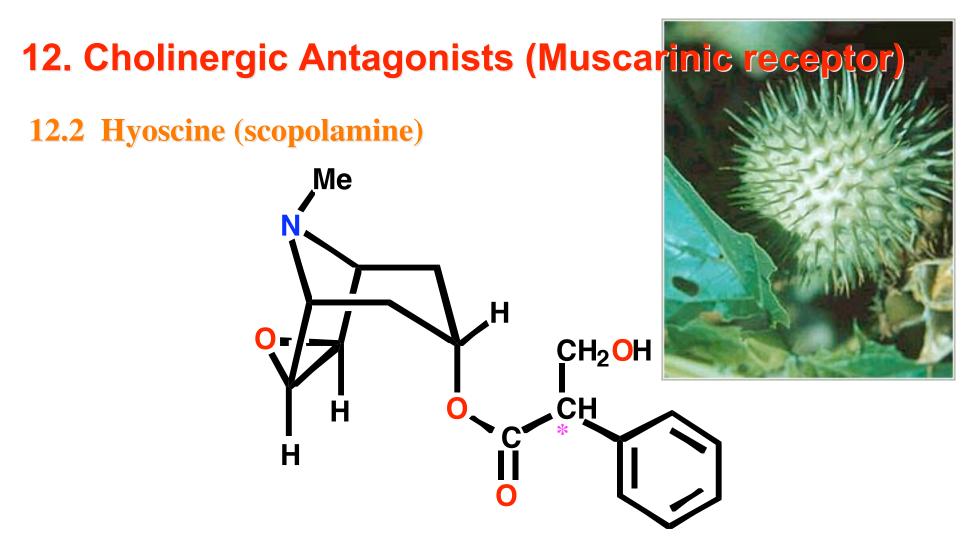
- Source roots of belladonna (1831) (deadly nightshade)
- Used as a poison
- Used as a medicine

decreases GIT motility antidote for anticholinesterase poisoning dilation of eye pupils

CNS side effects - hallucinations

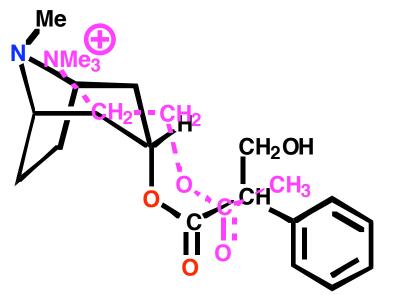
www.naturfoto.cz

CH₂OH



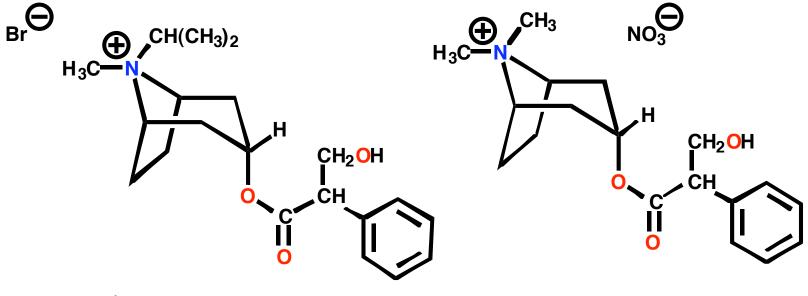
- Source thorn apple-*Datura*-jimsonweed
- Medical use treatment of motion sickness
- CNS effects, hallucinations

12.3 Comparison of atropine with acetylcholine



- Relative positions of ester and nitrogen similar in both molecules
- Nitrogen in atropine is ionised
- Amine and ester are important binding groups (ionic + H-bonds)
- Aromatic ring of atropine is an extra binding group (vdW)
- Atropine binds with a different induced fit no activation
- Atropine binds more strongly than acetylcholine

12.4 Analogues of atropine

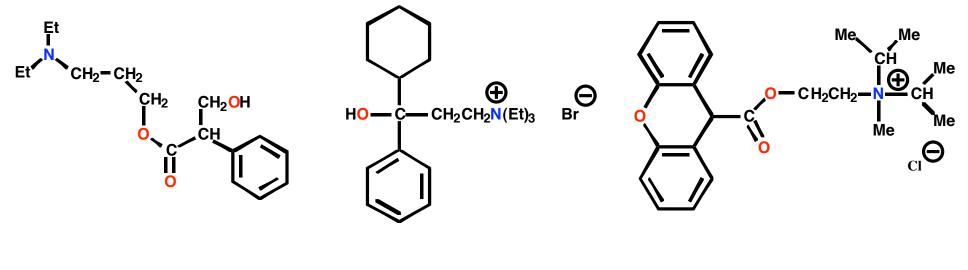


Ipratropium (bronchodilator & anti-asthmatic) Atropine methonitrate (lowers GIT motility)

- Analogues are fully ionized
- Analogues unable to cross the blood brain barrier
- No CNS side effects

12.5 Simplified Analogues

Pharmacophore = ester + basic amine + aromatic ring

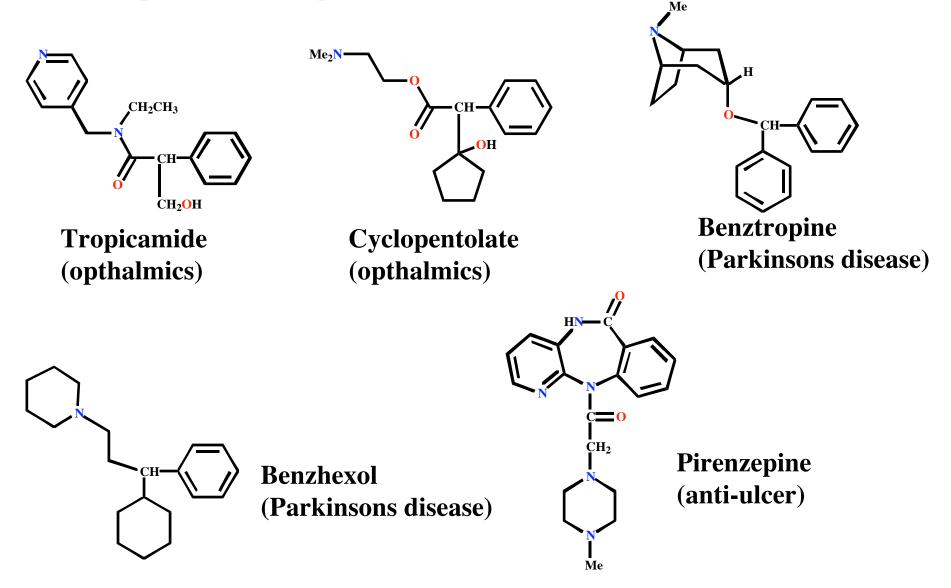


Amprotropine

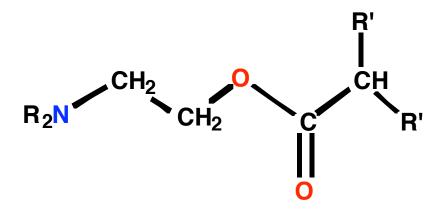
Tridihexethyl bromide

Propantheline chloride

12.5 Simplified Analogues



12.6 SAR for Antagonists

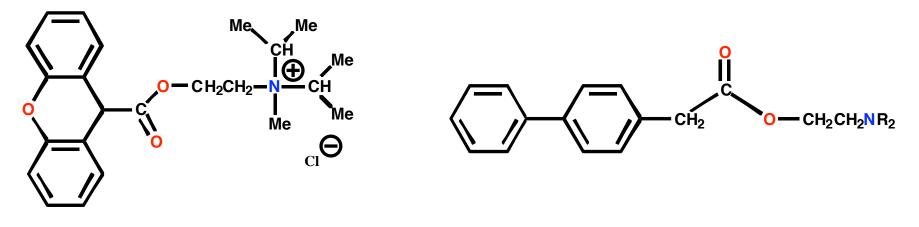


R' = Aromatic or Heteroaromatic

Important features

- Tertiary amine (ionised) or a quaternary nitrogen
- Aromatic ring
- Ester
- *N*-Alkyl groups (R) can be larger than methyl (unlike agonists)
- Large branched acyl group
- **R'** = aromatic or heteroaromatic ring
- Branching of aromatic/heteroaromatic rings is important

12.6 SAR for Antagonists



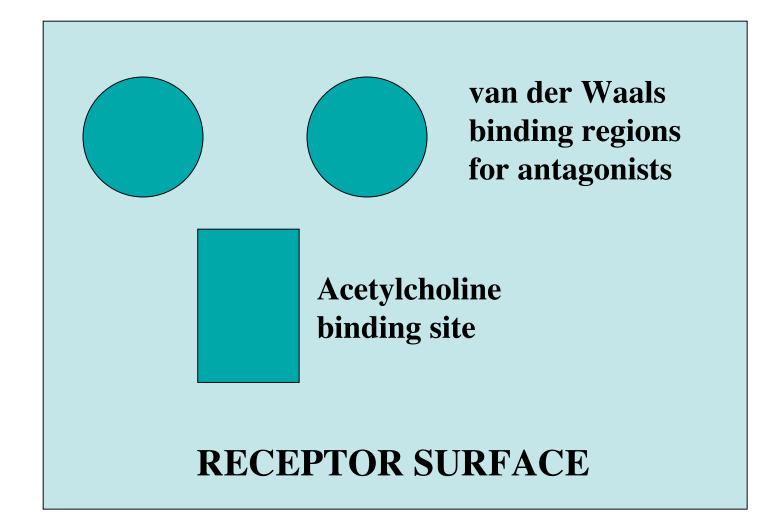
Active

Inactive

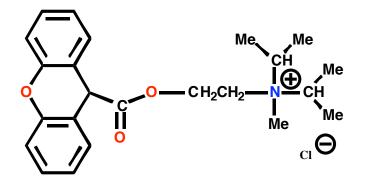
12.6 SAR for Antagonists vs. Agonists

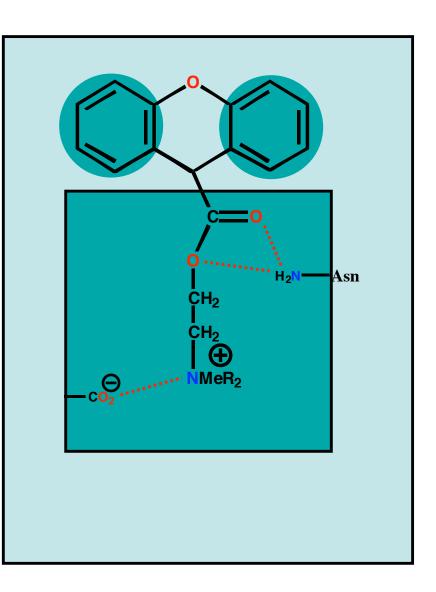
SAR for Antagonists	SAR for Agonists
Tertiary amine (ionized)	Quaternary nitrogen
or quaternary nitrogen Aromatic ring Ester <i>N</i> -Alkyl groups (R) can be larger than methyl R' = aromatic or heteroaromatic Branching of Ar rings important	Aromatic ring Ester N-Alkyl groups = methyl R' = H

12.7 Binding Site for Antagonists



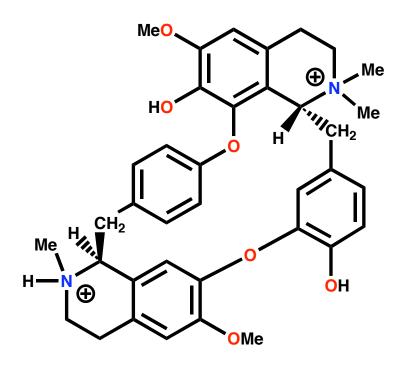
12.7 Binding Site for Antagonists





13.1 Curare

- Extract from curare plant- Strychnos toxifera
- Used for poison arrows
- Causes paralysis (blocks acetylcholine signals to muscles)
- Active principle = tubocurarine



Tubocurarine

CURARE

Strychnas toxifers Beetham

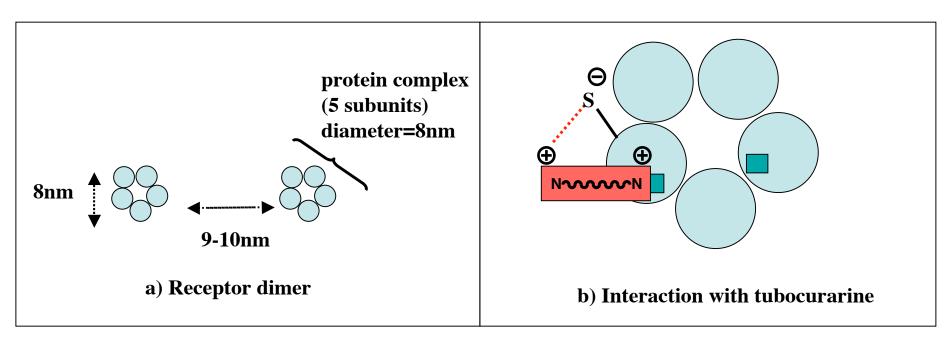
Pharmacophore

- Two quaternary centres at specific separation (1.15nm)
- Different mechanism of action from atropine based antagonists
- Different binding interactions

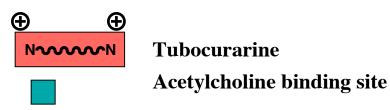
Clinical uses

- Neuromuscular blocker for surgical operations
- Permits lower and safer levels of general anaesthetic
- Tubocurarine used (previously) as neuromuscular blocker but side effects

13.2 Binding

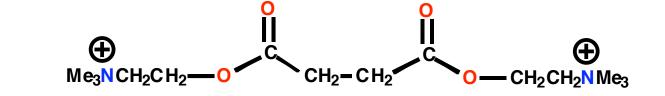


Probably not bridging Ach sites



13.3 Analogues of tubocurarine

(+)



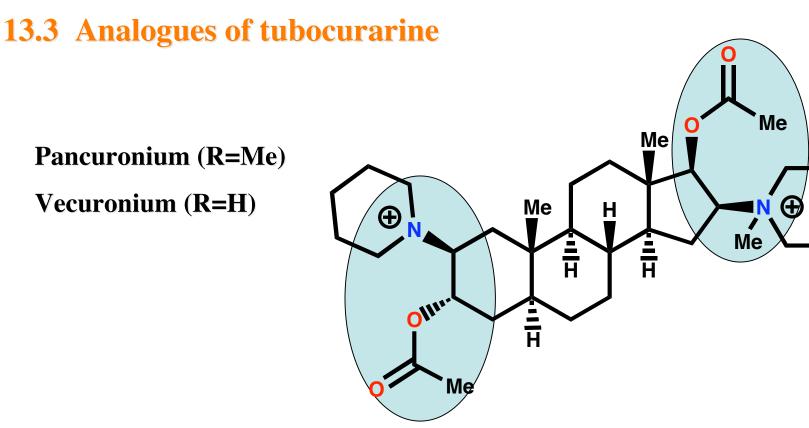
Decamethonium

Me₃N(CH₂)₁₀NMe₃

Ð

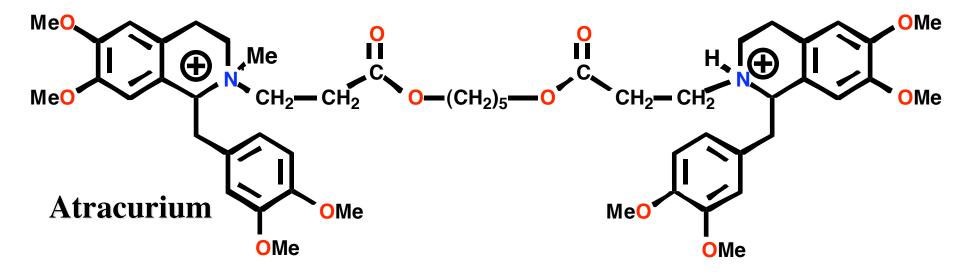
Suxamethonium

- Long lasting
- Long recovery times
- Side effects on heart
- Esters incorporated
- Shorter lifetime (5 min)
- Fast onset and short duration
- Side effects at autonomic ganglia



- Steroid acts as a spacer for the quaternary centres (1.09nm)
- Acyl groups are added to introduce the Ach skeleton
- Faster onset then tubocurarine but slower than suxamethonium
- Longer duration of action than suxamethonium (45 min)
- No effect on blood pressure and fewer side effects

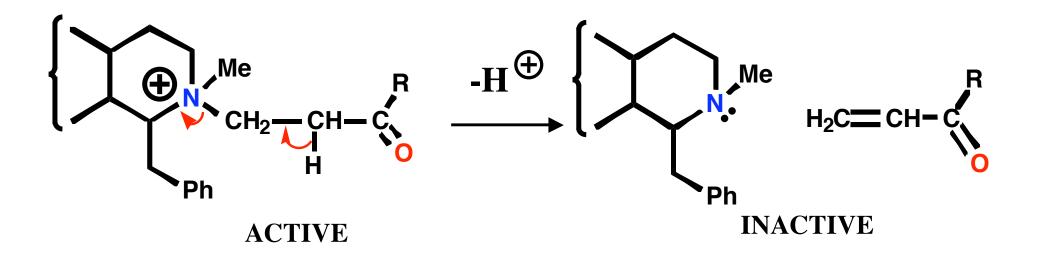
13.3 Analogues of tubocurarine



- Design based on tubocurarine and suxamethonium
- Lacks cardiac side effects
- Rapidly broken down in blood both chemically and metabolically
- Avoids patient variation in metabolic enzymes
- Lifetime is 30 minutes
- Administered as an i.v. drip
- Self destruct system limits lifetime

13. Cholinergic Antagonists (Nicotinic receptor)

13.3 Analogues of tubocurarine

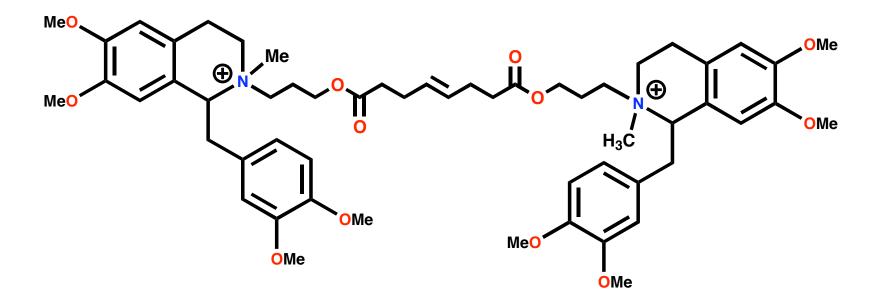


Atracurium stable at acid pH Hofmann elimination at blood pH (7.4)

13. Cholinergic Antagonists (Nicotinic receptor)

13.3 Analogues of tubocurarine

Mivacurium



- Faster onset (2 min)
- Shorter duration (15 min)

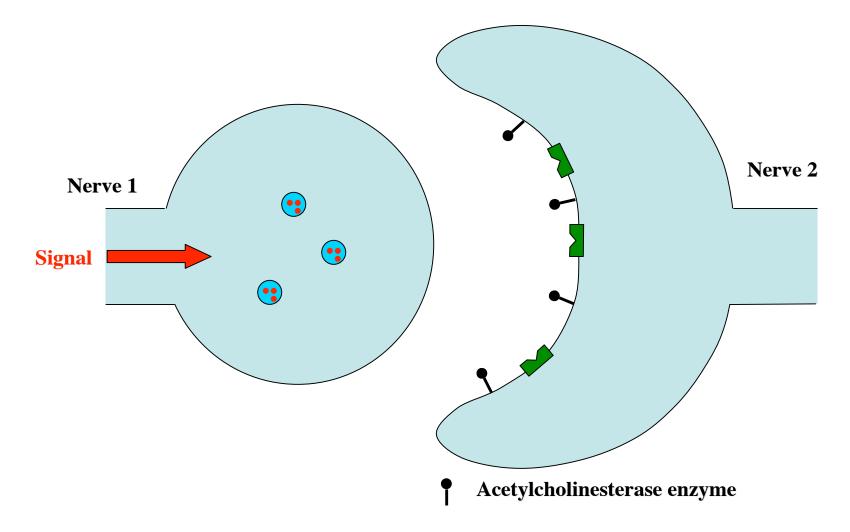
Contents

Part 3: Cholinergics & anticholinesterases

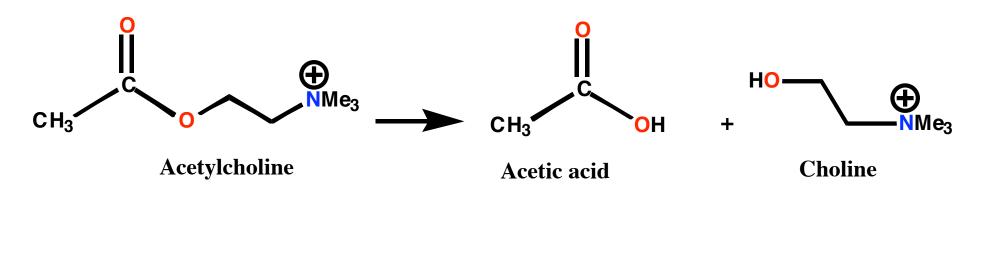
- 14. Acetylcholinesterase
 - 14.1. Role
 - 14.2. Hydrolysis reaction catalysed
 - 14.3. Effect of inhibition
 - 14.4. Structure of enzyme complex
 - 14.5. Active site binding interactions
 - 14.6. Active site Mechanism of catalysis
- 15. Anticholinesterases
 - 15.1. Physostigmine
 - 15.2. Mechanism of action
 - 15.3. Physostigmine analogues
 - 15.4. Organophosphates
 - 15.5. Anticholinesterases as 'Smart Drugs'

14.1 Role

- Hydrolysis and deactivation of acetylcholine
- Prevents acetylcholine reactivating receptor



14.2 Hydrolysis reaction catalysed

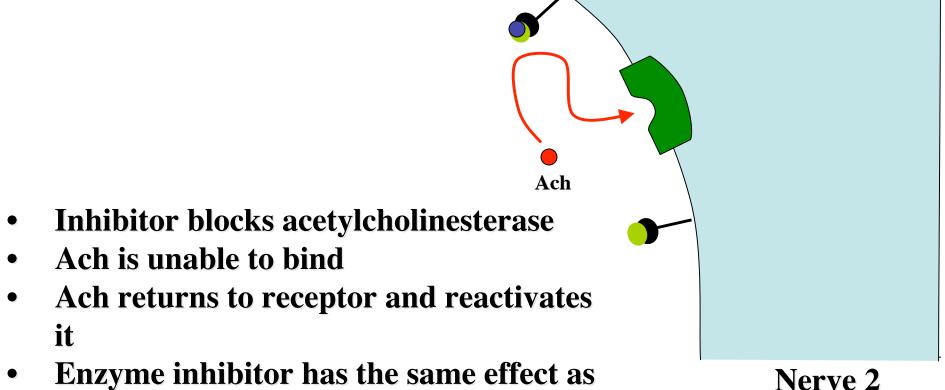


active

inactive

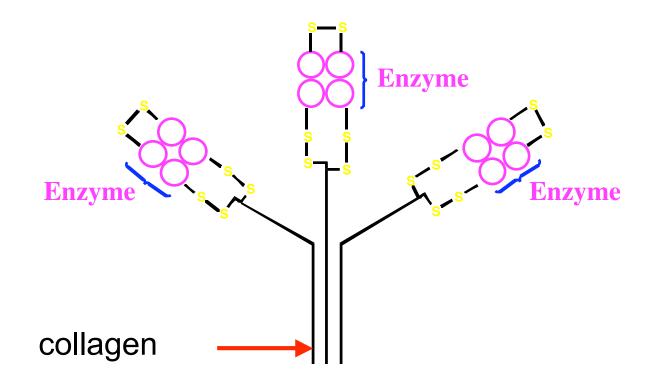
14.3 Effect of inhibition

Enzyme inhibitor (Anticholinesterase)

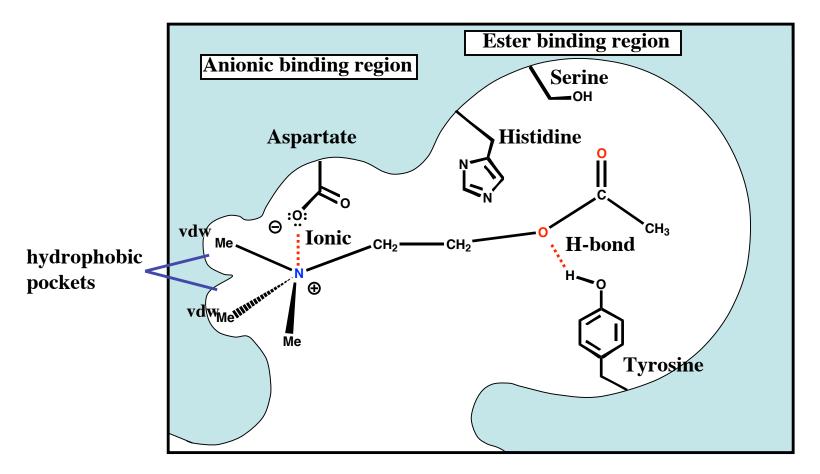


• Enzyme inhibitor has the same effect as a cholinergic agonist

14.4 Structure of enzyme complex

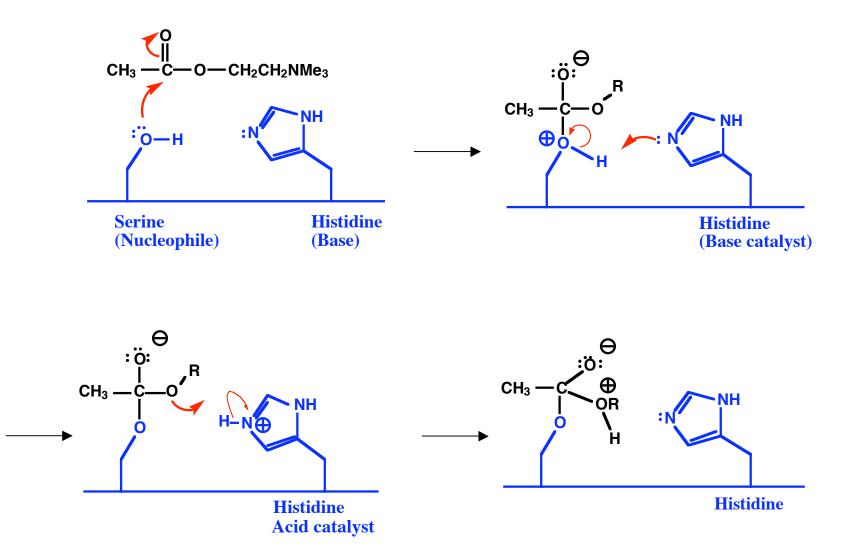


14.5 Active site - binding interactions

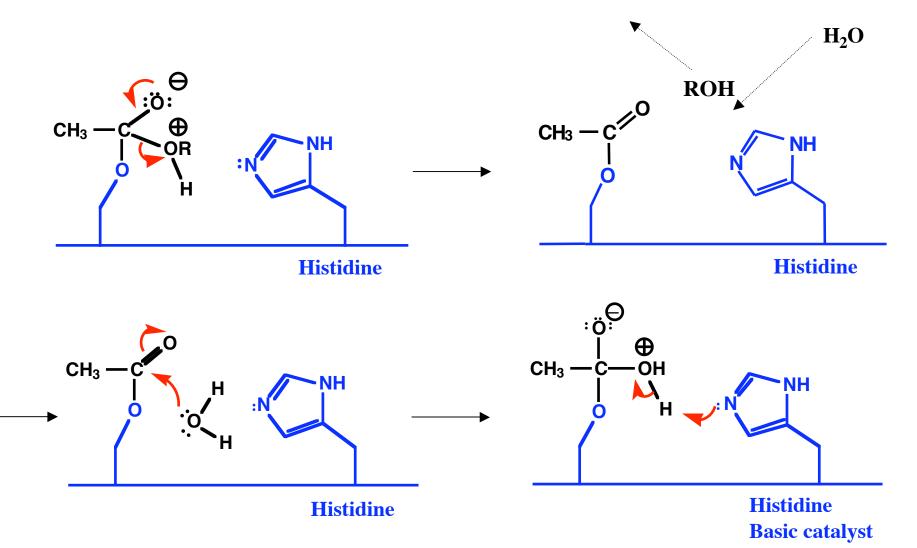


- Anionic binding region similar to cholinergic receptor site
- Binding and induced fit strains Ach and weakens bonds
- Molecule positioned for reaction with His and Ser

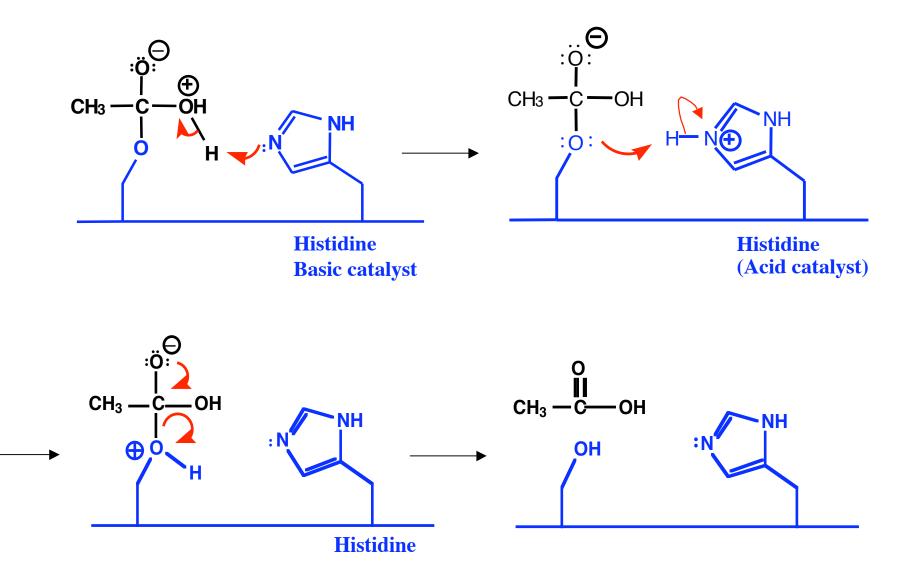
14.6 Active site - Mechanism of catalysis



14.6 Active site - Mechanism of catalysis



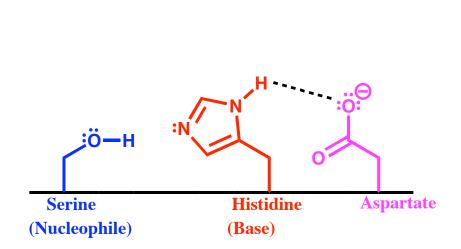
14.6 Active site - Mechanism of catalysis



- Serine and water are poor nucleophiles
- Mechanism is aided by histidine acting as a basic catalyst
- Choline and serine are poor leaving groups
- Leaving groups are aided by histidine acting as an acid catalyst
- Very efficient 100 x 10⁶ faster than uncatalysed hydrolysis
- Acetylcholine hydrolysed within 100 µsecs of reaching active site
- An aspartate residue is also involved in the mechanism

The catalytic triad

• An aspartate residue interacts with the imidazole ring of histidine to orient and activate it

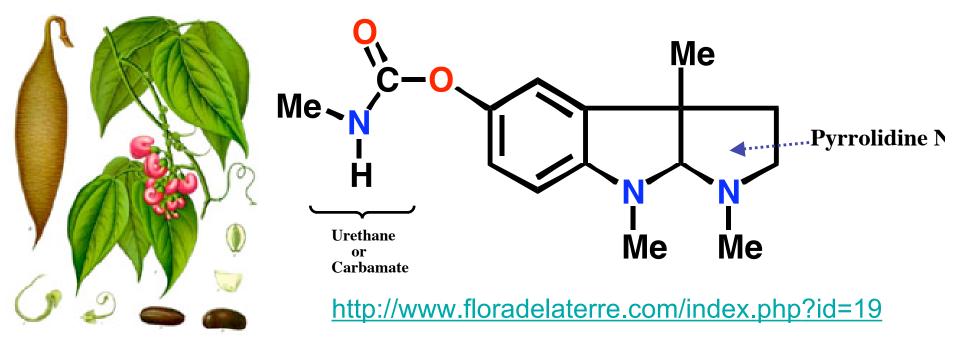


15. Anticholinesterases

- Inhibitors of acetylcholinesterase enzyme
- Block hydrolysis of acetylcholine
- Acetylcholine is able to reactivate cholinergic receptor
- Same effect as a cholinergic agonist

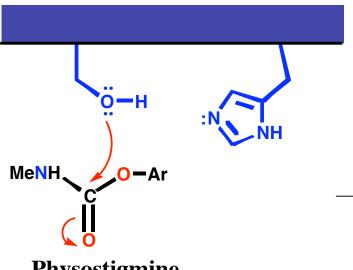
15. Anticholinesterases

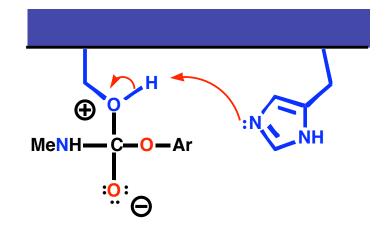
15.1 Physostigmine



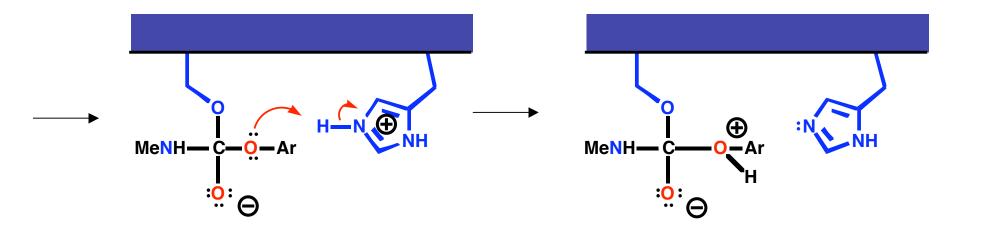
- Natural product from the African calabar (ordeal) bean-Physostigma
- Carbamate is essential (equivalent to ester of Ach)
- Aromatic ring is important
- Pyrrolidine N is important (ionized at blood pH)
- Pyrrolidine N is equivalent to the quaternary nitrogen of Ach

15.2 Mechanism of action

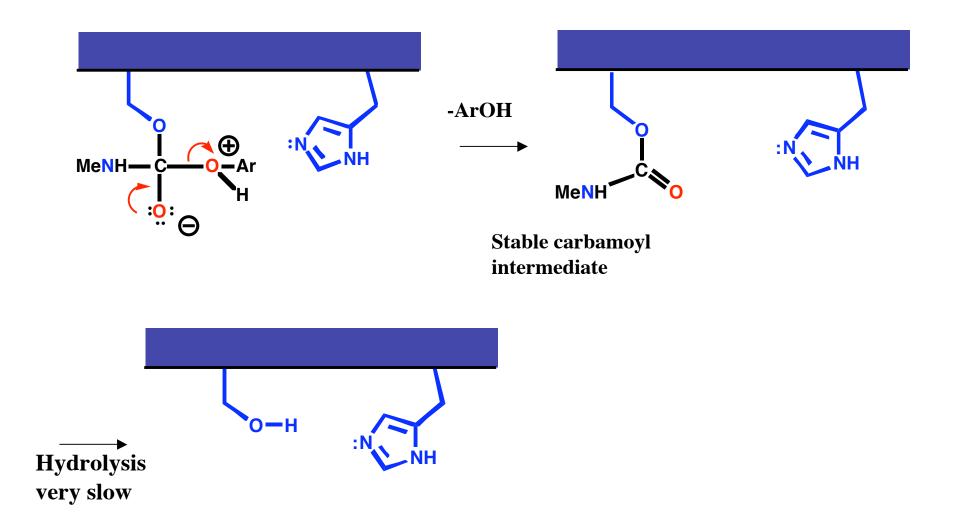




Physostigmine

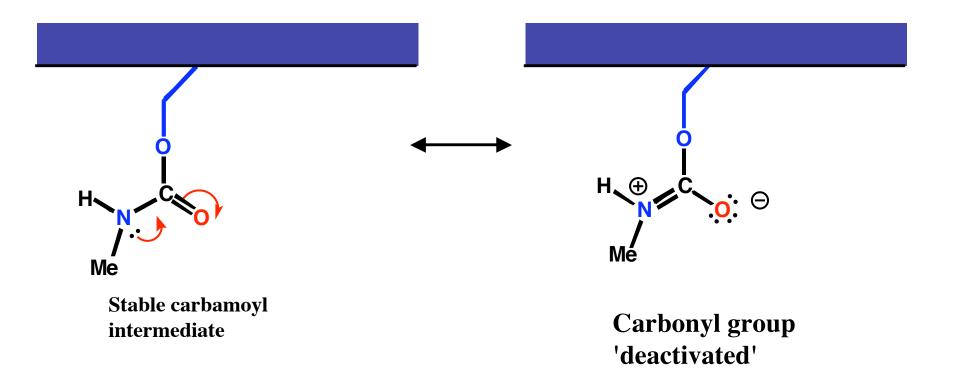


15.2 Mechanism of action

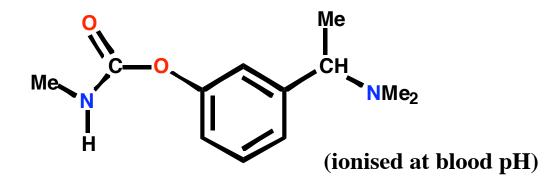


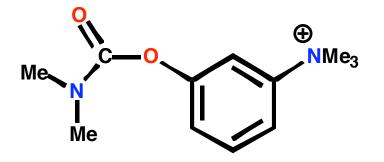
Rate of hydrolysis slower by 40 x 10⁶

15.2 Mechanism of action



15.3 Physostigmine analogues





Miotine

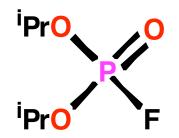
- Simplified analogue
- Susceptible to hydrolysis
- Crosses BBB as free base
- CNS side effects

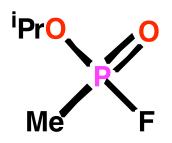
Neostigmine

- Fully ionized
- Cannot cross BBB
- No CNS side effects
- More stable to hydrolysis
- Extra *N*-methyl group increases stability



a) Nerve gases



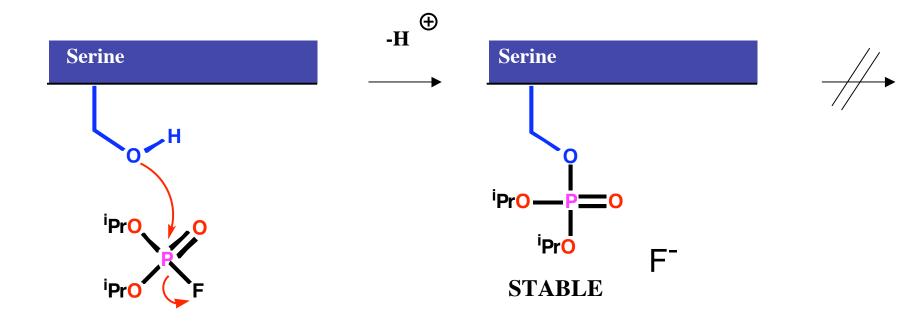


Sarin



- Agents developed in World War 2
- Agents irreversibly inhibit acetylcholinesterase
- Permanent activation of cholinergic receptors by Ach
- Results in death

b) Mechanism of action

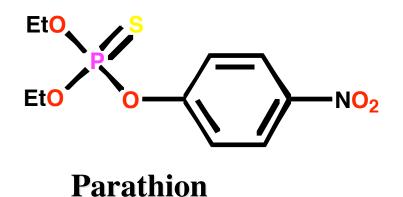


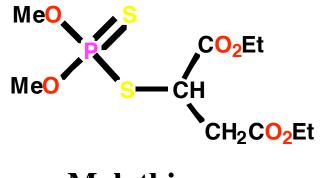
- Irreversible phosphorylation
- P-O bond very stable

c) Medicinal organophosphate

- Used to treat glaucoma
- Topical application
- Quaternary N is added to improve binding interactions
- Results in better selectivity and lower, safer doses

d) Organophosphates as insecticides



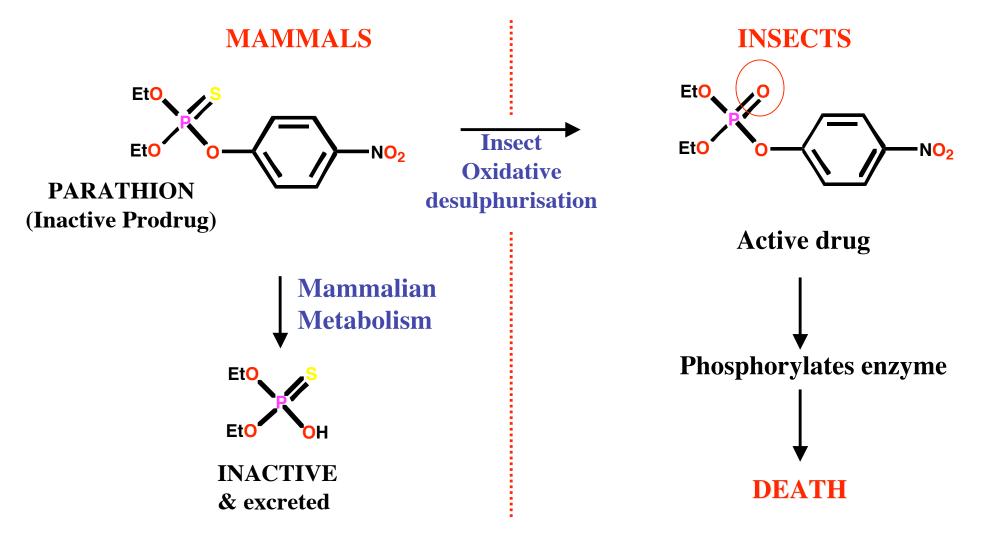


Malathion

- Relatively harmless to mammals
- Agents act as prodrugs in insects
- Metabolised by insects to produce a toxic metabolite



d) Organophosphates as insecticides



e) Design of Organophosphate Antidotes

Strategy

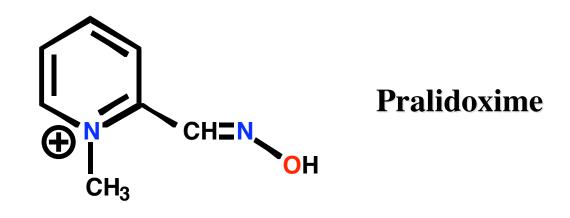
- Strong nucleophile required to cleave strong P-O bond
- Find suitable nucleophile capable of cleaving phosphate esters
- Water is too weak as a nucleophile
- Hydoxylamine is a stronger nucleophile

$$NH_{2}OH + RO - P - OR \longrightarrow 0$$

$$Hydroxylamine OR + ROH$$

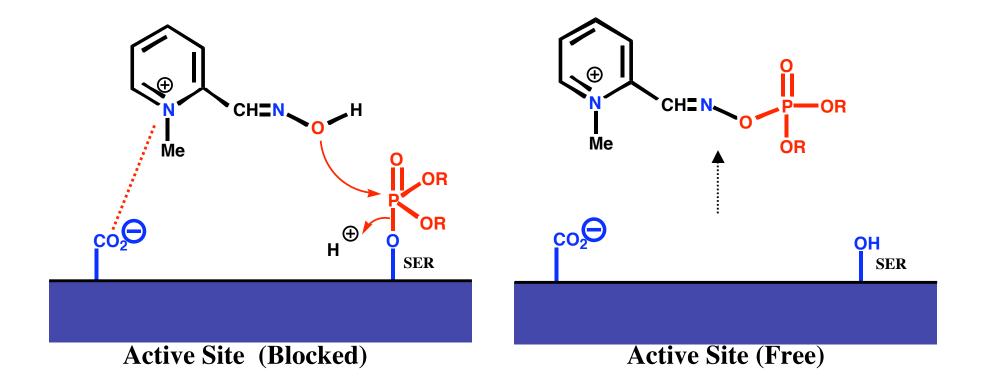
- Hydroxylamine is too toxic for clinical use
- Increase selectivity by increasing binding interactions with active site

e) Design of Organophosphate Antidotes

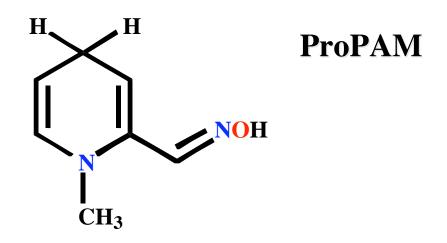


- Quaternary N is added to bind to the anionic region
- Side chain is designed to place the hydroxylamine moiety in the correct position relative to phosphorylated serine
- Pralidoxime 1 million times more effective than hydroxylamine
- Cannot act in CNS due to charge cannot cross bbb

e) Design of Organophosphate Antidotes



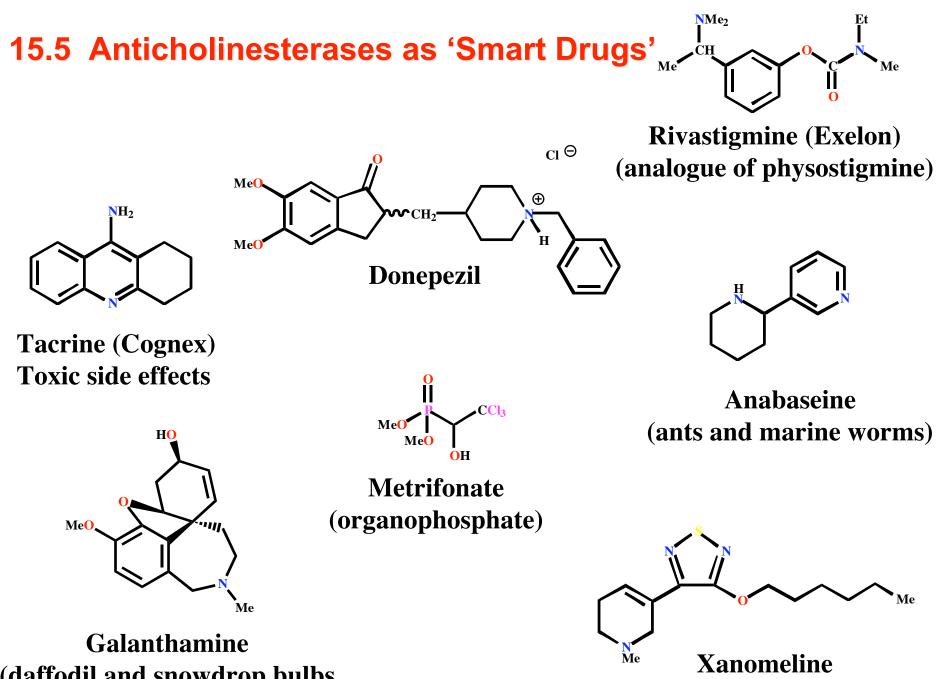
e) Design of Organophosphate Antidotes



- Prodrug for pralidoxime
- Passes through BBB as free base
- Oxidised in CNS to pralidoxime

15.5 Anticholinesterases as 'Smart Drugs'

- Act in CNS
- Must cross blood brain barrier
- Used to treat memory loss in Alzheimers disease
- Alzheimers causes deterioration of cholinergic receptors in brain
- Smart drugs inhibit Ach hydrolysis to increase activity at remaining receptors



(daffodil and snowdrop bulbs