

Chem 352 - Spring 2011

Quiz 4

1. The word “gluconeogenesis” translates as “to create new sugar”.
 - a. In mammals, what organ is the primary site for gluconeogenesis? _____
 - b. What is the primary purpose for having this organ synthesize new glucose? _____

- c. The final reaction in glycolysis is catalyzed by the enzyme *pyruvate kinase* and converts phosphoenolpyruvate to pyruvate. Under physiological conditions, this reaction has a very high negative free energy change, therefore, converting pyruvate back into phosphoenolpyruvate in gluconeogenesis requires an alternative pathway. Using *structural formulas* for the reactants, products and intermediates, write the *balance chemical equations* for the two reactions in this pathway.

- d. What are the names of the enzymes that catalyze the two reactions described in c?
 - i. _____
 - ii. _____
- e. What is the sources of free energy used to drive this reaction? _____
- f. What citric acid cycle intermediate is also involved as an intermediate in the conversion of pyruvate to phosphoenolpyruvate? _____

2. In its catabolic mode, the citric acid cycle is used to oxidize the equivalent of the two carbons that enter the cycle as acetyl-CoA and releases them as CO₂.
- a. Using *structural formulas* for the reactants and products, write the balanced chemical equation for *one* of the oxidoreductase reactions in the citric acid cycle:
- b. What is the name of the enzyme that catalyzes the reaction you chose in a? _____
3. The oxidative stage of the pentose phosphate pathway is used to convert a glycolytic intermediate to a pentose and at the same time produce reducing agents for biosynthetic reactions.
- a. Which glycolytic intermediate is the starting point for the oxidative stage of the pentose phosphate pathway? _____
- b. What is the name of the pentose produced? _____
- c. What reducing agents are formed? _____
- d. Besides two oxidoreductase reactions, the oxidative stage also contains a hydrolase reaction. Using *structural formulas* for the reactant and product, write the *balanced chemical equation* for this reaction.
- e. If the pentose produced in the pentose phosphate pathway is not required by the cell, what two glycolytic intermediates is it converted to by way of the non-oxidative stage of the pentose phosphate pathway?
- i. _____
- ii. _____