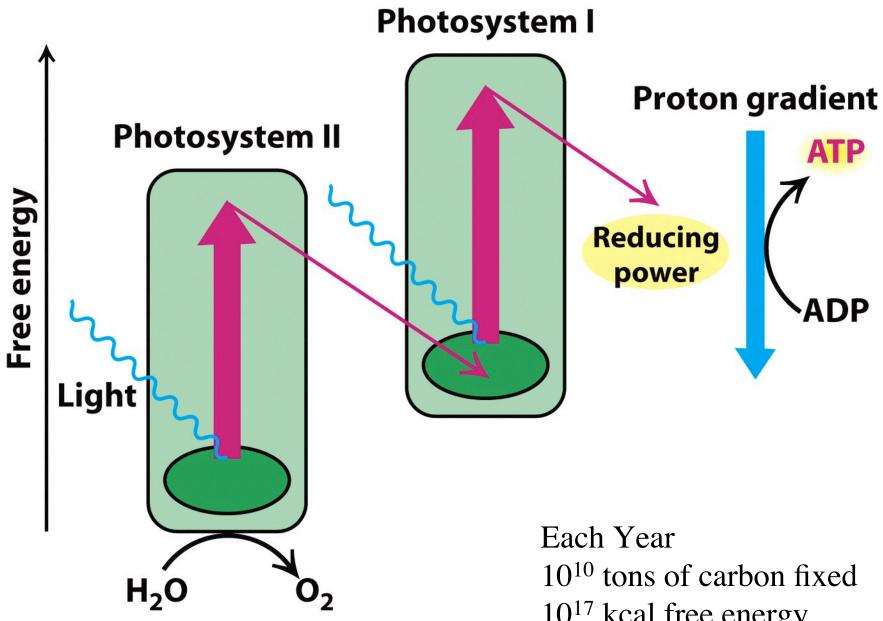
A Fast Run Through Plants

Chapter 19

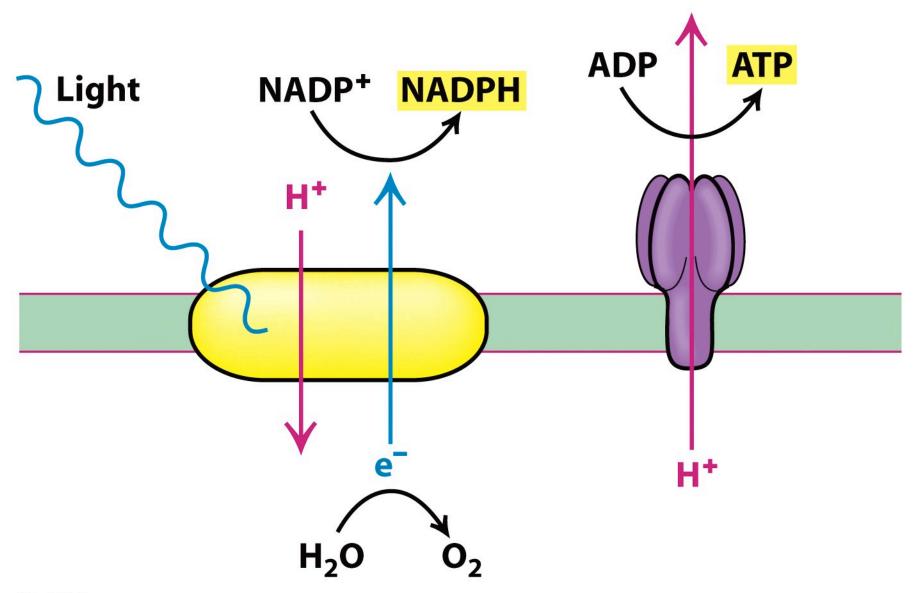
The Light Reactions of Photosynthesis

An AOL Time Warper Company



Chapter 19 Opener part 2 **Biochemistry, Sixth Edition** © 2007 W. H. Freeman and Company 10¹⁷ kcal free energy

The Light Reactions



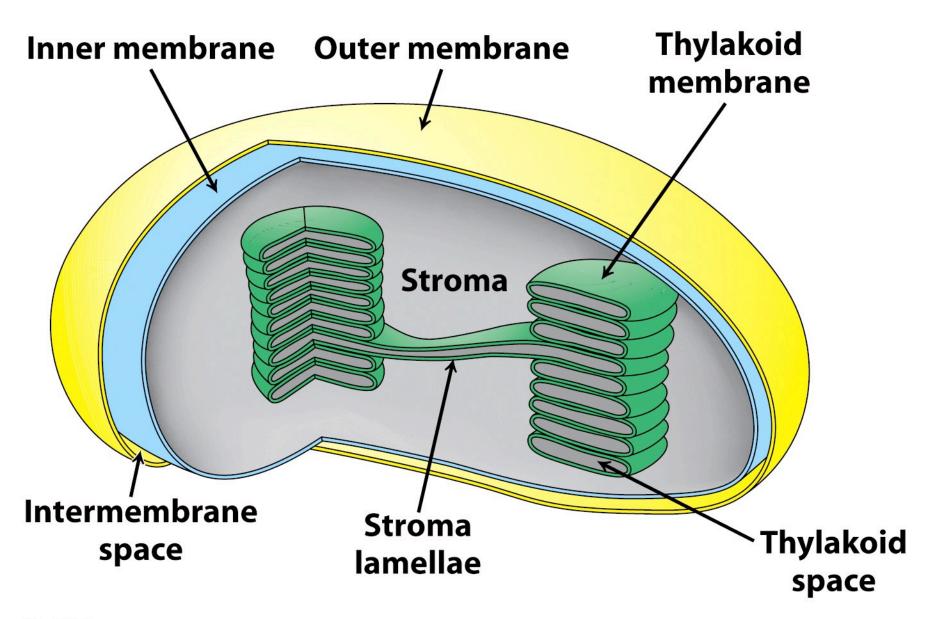


Figure 19-3 Biochemistry, Sixth Edition © 2007 W.H.Freeman and Company

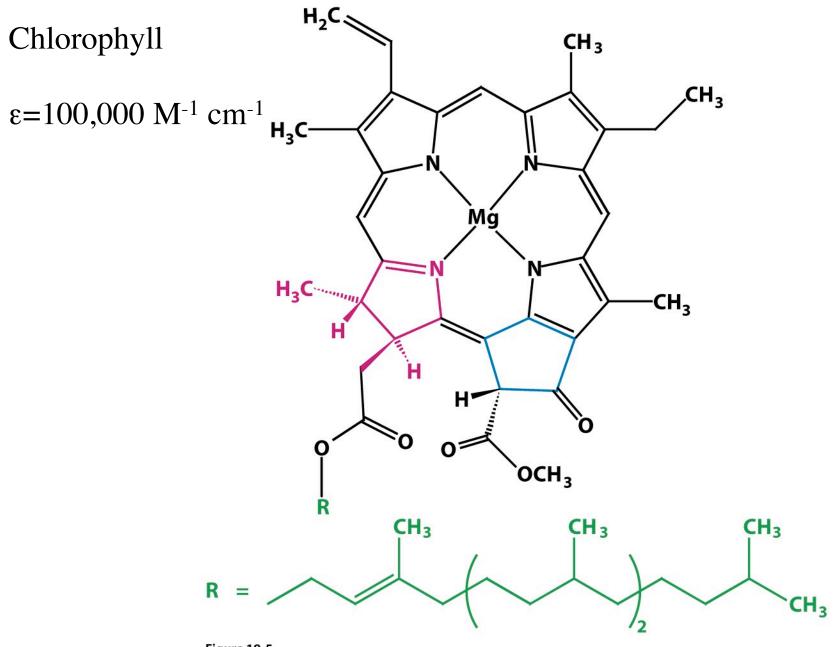


Figure 19-5 Biochemistry, Sixth Edition © 2007 W.H.Freeman and Company Two Chloroplast Photosystems

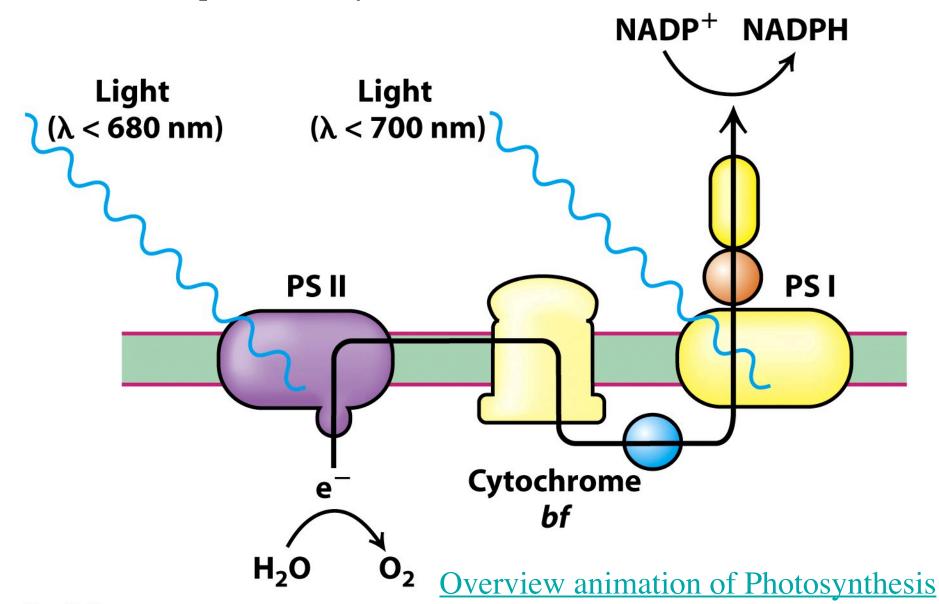
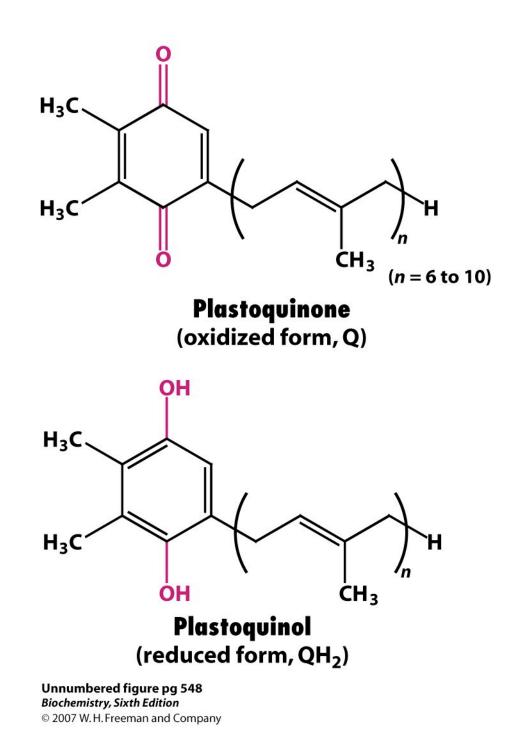
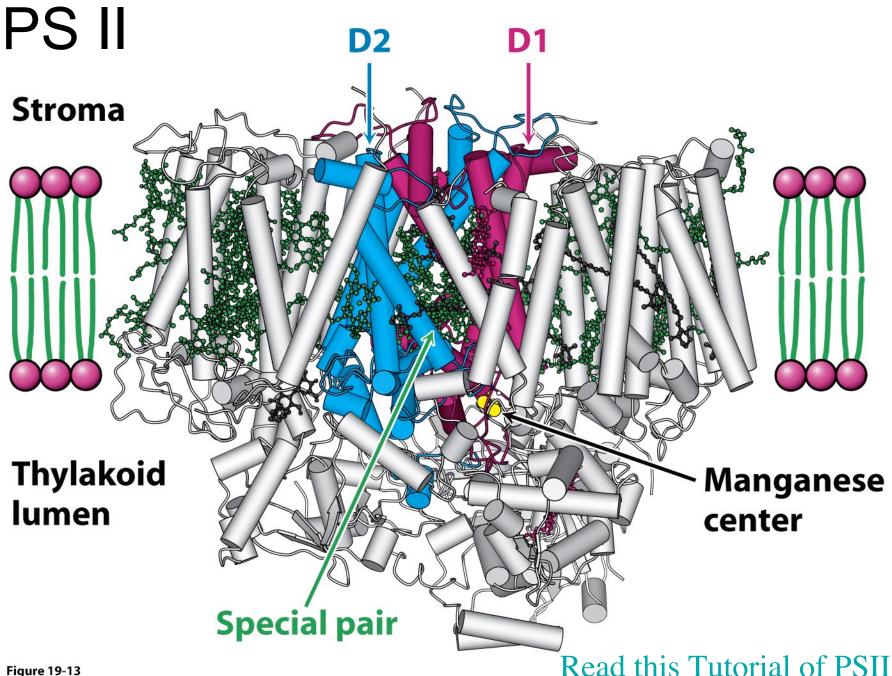


Figure 19-12 Biochemistry, Sixth Edition © 2007 W.H.Freeman and Company





Read this Tutorial of PSII

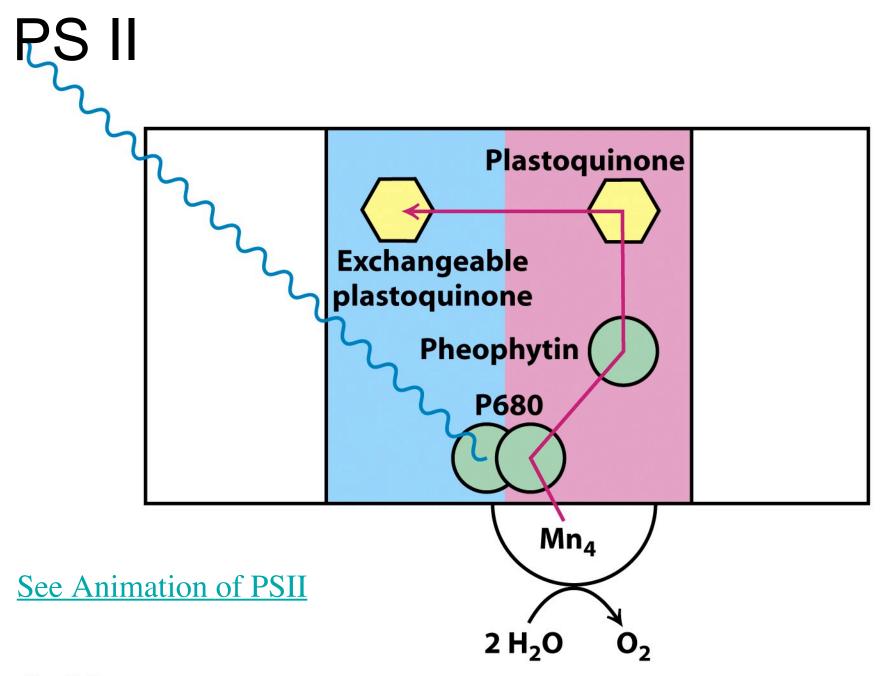


Figure 19-14 Biochemistry, Sixth Edition © 2007 W.H.Freeman and Company

Oxidizing Water! Each photon "pulls off" an electron

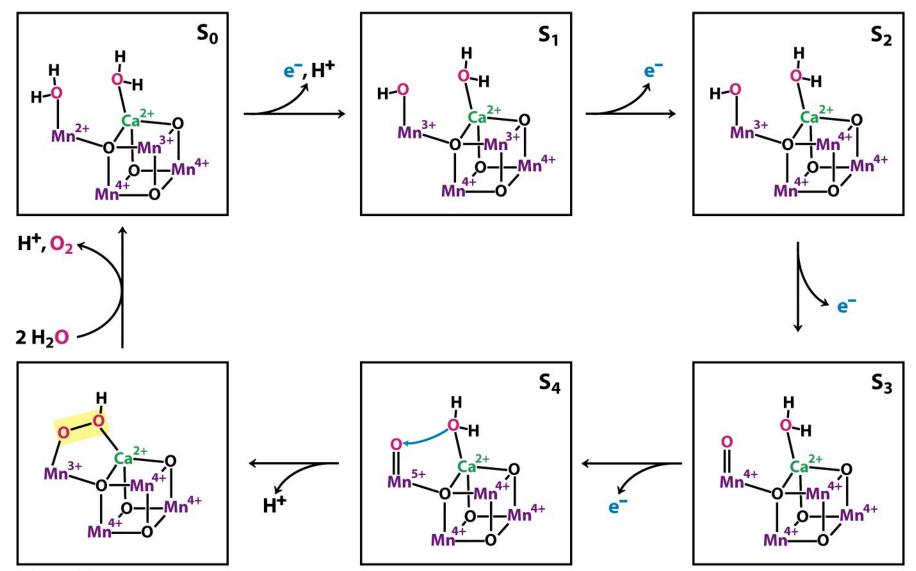


Figure 19-16 Biochemistry, Sixth Edition © 2007 W.H. Freeman and Company

PS II- generates H+ gradient too (Jagendorf, 1966) $4 H^{+}$ Stroma (high pH) 2 QH₂ 20 $O_2 + 4 H^+$ $2 H_{2}O$ **Thylakoid lumen** (low pH)

Figure 19-17 Biochemistry, Sixth Edition © 2007 W. H. Freeman and Company

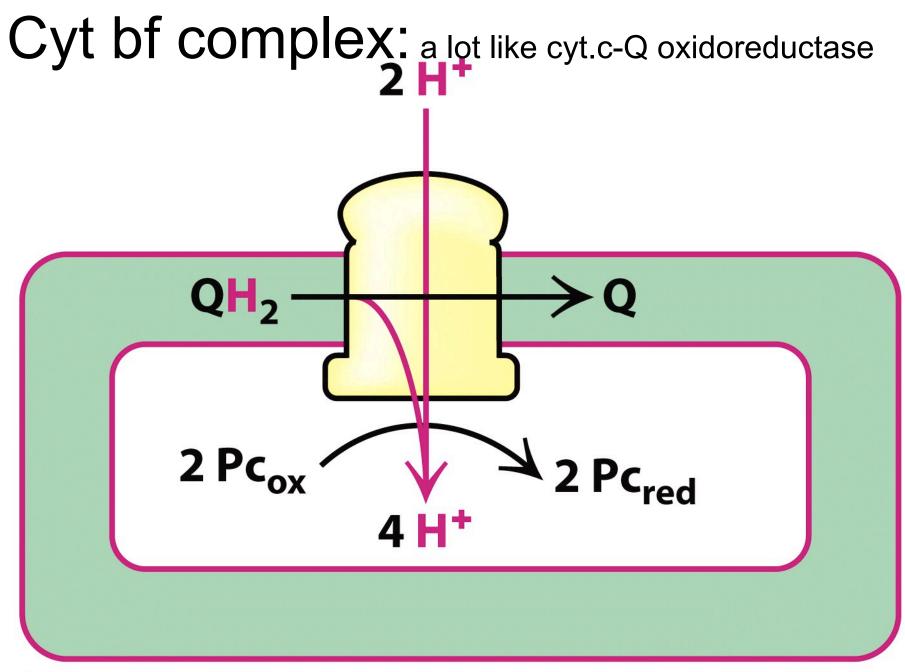


Figure 19-18 Biochemistry, Sixth Edition © 2007 W.H.Freeman and Company

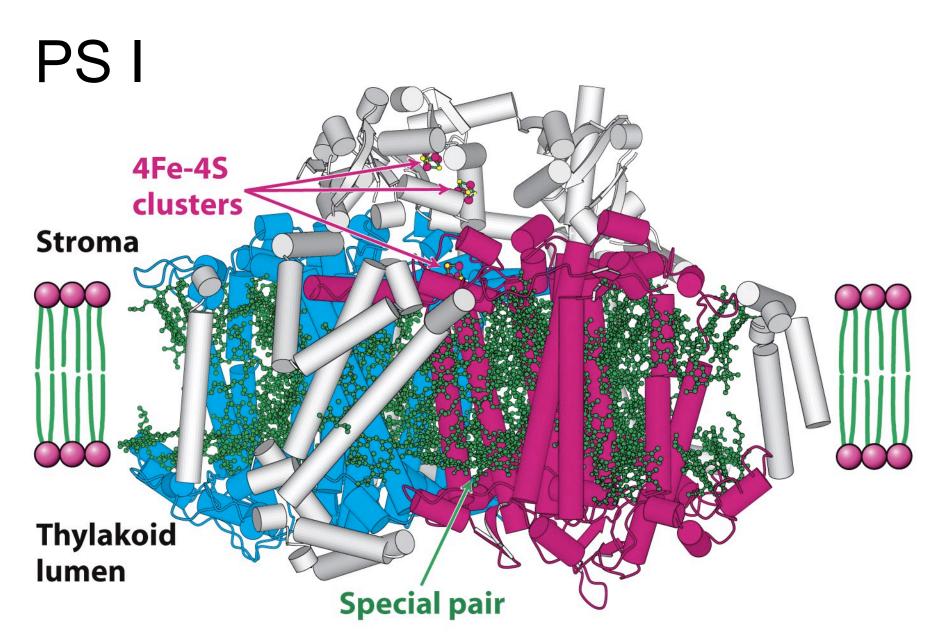
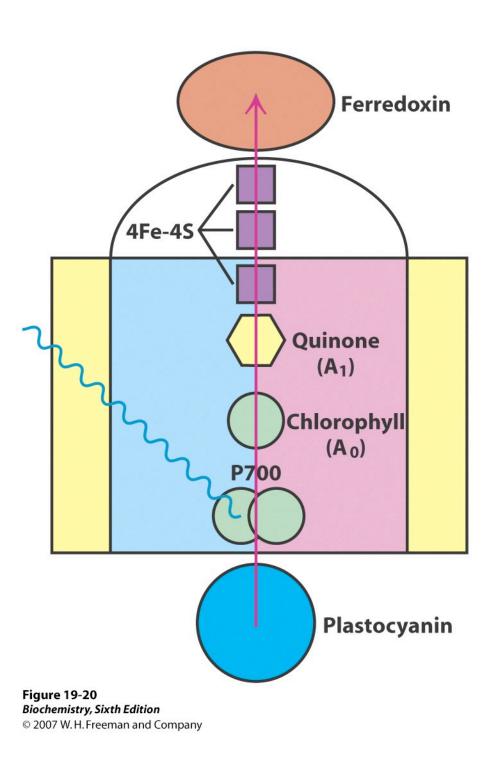
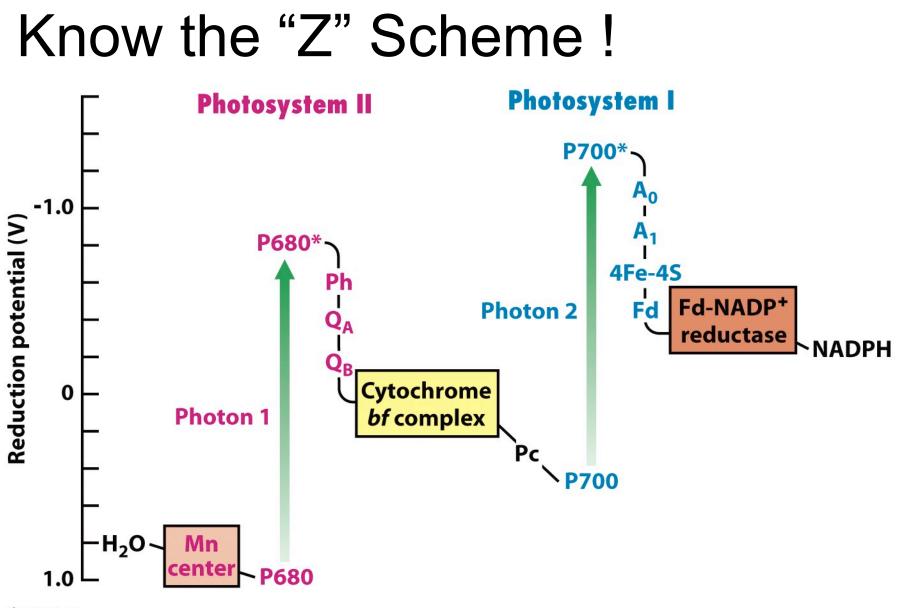


Figure 19-19 Biochemistry, Sixth Edition © 2007 W.H. Freeman and Company

Read this Tutorial of PSI







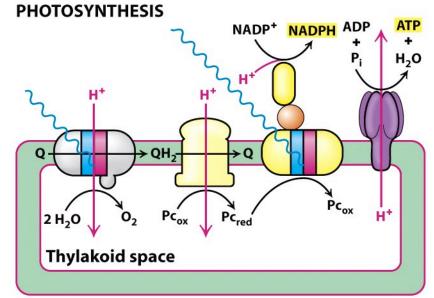


Overview animation of Photosynthesis

ENERGETICS

In chloroplasts $\Delta pH=3.5$

So, ΔG = - 4.8 kcal /mol H⁺ and there are ~4 H⁺/ ATP



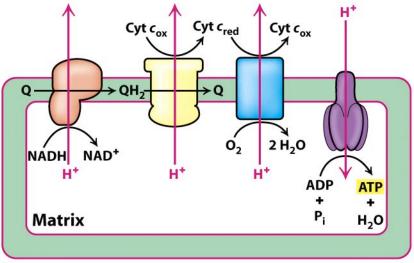
Stroma

OXIDATIVE PHOSPHORYLATION

$$\label{eq:G} \begin{split} \Delta G &= 2.3 \ \text{RT} \ (\text{pH}_{thylakoid} - \text{pH}_{stroma}) \\ &+ F \Delta \Psi_{(\text{out relative to in})} \end{split}$$

Overall Photosynthesis gives:

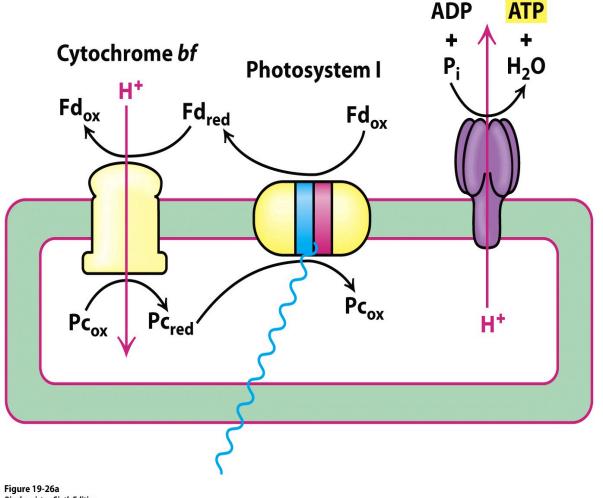
8 photons = $1O_2$, 2 NADPH, 3 ATP (a 700 nm photon has -40.9 kcal /mole potentially)



Intermembrane space

Figure 19-25 Biochemistry, Sixth Edition © 2007 W.H. Freeman and Company

Cyclic Photophosphorylation: for ATP, only when NADPH is high!



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