

Chem 452 - Lecture 8

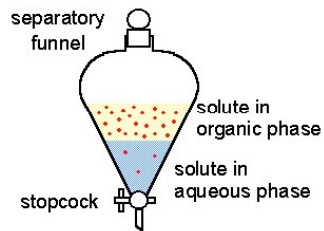
Lipids and Cell Membranes

Part 1

Question of the Day: What makes a lipid a lipid?

Introduction

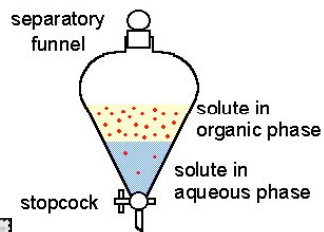
* **Lipids** are the components of a cell that can be extracted with organic solvents.



Chem 452, Lecture 8 - Lipids and Cell Membranes 2

Introduction

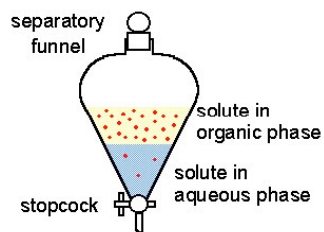
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Introduction

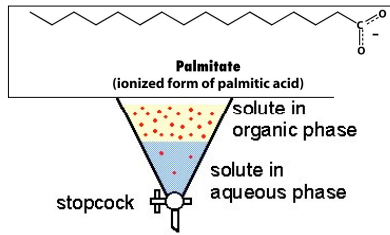
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Introduction

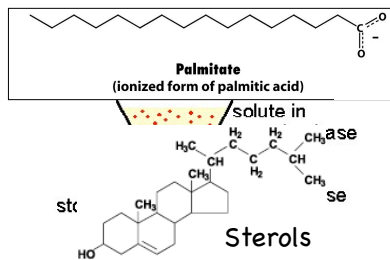
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Introduction

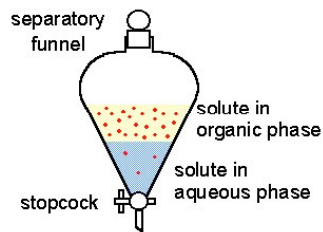
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Introduction

♦ **Lipids** are the components of a cell that can be extracted with organic solvents.



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Lipids

♦ Are the water-insoluble molecules found in a living cell.

♦ Roles for these molecules include

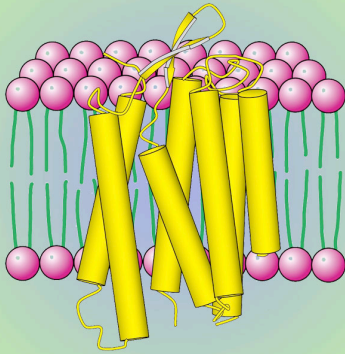
- Fuel
 - fatty acids and ketone bodies
- Long term storage
 - Triacylglycerides (fat)
- Messengers in signal transduction
 - Steroids
 - Diacylglycerol
- Components of membranes

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Introduction

- † Biological **membranes** define the boundary of a cell.
 - Cellular communications with the surroundings are mediated by cell membrane
- † “Membranes are dynamics structures in which proteins float in a sea of lipids.”

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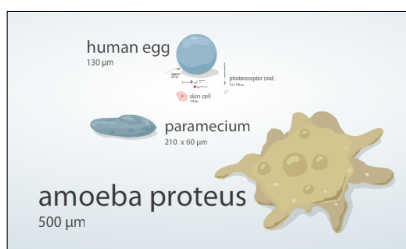
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Introduction

- † Membranes distinguish eukaryotic cells from prokaryotic cells.



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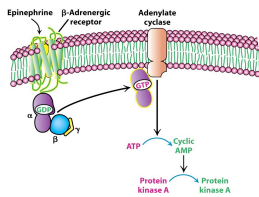
Introduction

- † Membranes play important roles.
 - Energy Storage
 - Information Transduction
 - Compartmentalization
- † Membranes are hydrophobic environments that separate two hydrophilic environments.

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Regulation by Covalent Modification

- † **Protein Kinase A (PKA)** is involved in the "flight or fight" response.
 - This response is triggered by the release of the hormone epinephrine (adrenalin) by the adrenal glands.

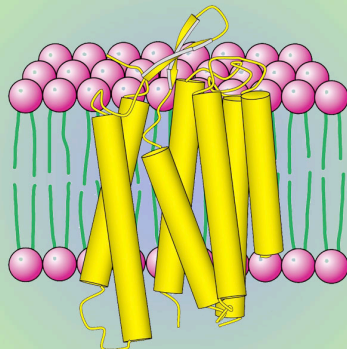


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Introduction

- † Features of membranes:
 - They are sheet-like structures that form compartments.
 - They are composed of lipids and proteins.
 - Membrane lipids are **amphipathic** (both hydrophilic and hydrophobic) and form the sheet-like structure
 - The proteins serve as the pumps, channels, receptors, energy transducers and enzymes.
 - Membranes are non-covalent, fluid, asymmetric, assemblies
 - Membranes are electrically polarized.

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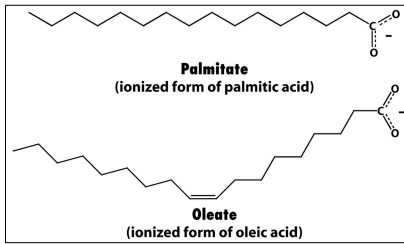
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Fatty Acids

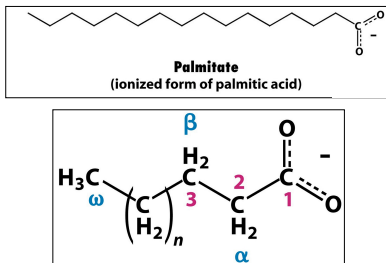
- Fatty acids are one of the major groups of lipids.



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Fatty Acids

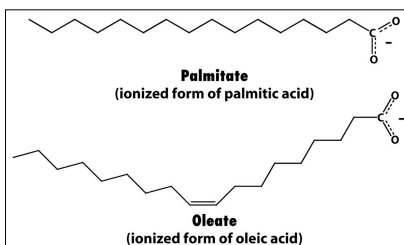
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Fatty Acids

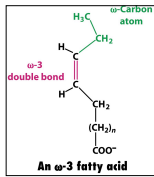
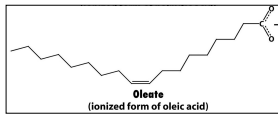
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Fatty Acids

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Fatty Acids

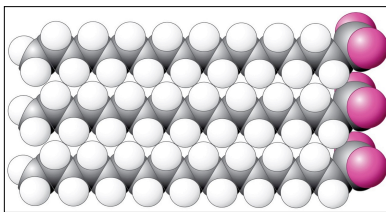
TABLE 12.1 Some naturally occurring fatty acids in animals

Number of carbons	Number of double bonds	Common name	Systematic name	Formula
12	0	Laurate	<i>n</i> -Dodecanoate	$\text{CH}_3(\text{CH}_2)_{10}\text{COO}^-$
14	0	Myristate	<i>n</i> -Tetradecanoate	$\text{CH}_3(\text{CH}_2)_{12}\text{COO}^-$
16	0	Palmitate	<i>n</i> -Hexadecanoate	$\text{CH}_3(\text{CH}_2)_{14}\text{COO}^-$
18	0	Stearate	<i>n</i> -Octadecanoate	$\text{CH}_3(\text{CH}_2)_{16}\text{COO}^-$
20	0	Arachidate	<i>n</i> -Eicosanoate	$\text{CH}_3(\text{CH}_2)_{18}\text{COO}^-$
22	0	Behenate	<i>n</i> -Docosanoate	$\text{CH}_3(\text{CH}_2)_{20}\text{COO}^-$
24	0	Lignocerate	<i>n</i> -Tetracosanoate	$\text{CH}_3(\text{CH}_2)_{22}\text{COO}^-$
16	1	Palmitoleate	<i>cis</i> - Δ^7 -Hexadecenoate	$\text{CH}_3(\text{CH}_2)_5\text{CH}=\text{CH}(\text{CH}_2)_9\text{COO}^-$
18	1	Oleate	<i>cis</i> - Δ^9 -Octadecenoate	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_9\text{COO}^-$
18	2	Linoleate	<i>cis</i> - Δ^9, Δ^{12} -Octadecadienoate	$\text{CH}_3(\text{CH}_2)_4\text{CH}=\text{CH}(\text{CH}_2)_4\text{CH}=\text{CH}(\text{CH}_2)_7\text{COO}^-$
18	3	Linolenate	all- <i>cis</i> - $\Delta^9, \Delta^{12}, \Delta^{15}$ -Octadecatrienoate	$\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}_2\text{CH}_2\text{COO}^-$
20	4	Arachidonate	all- <i>cis</i> - $\Delta^9, \Delta^{11}, \Delta^{14}, \Delta^{17}$ -Eicosatetraenoate	$\text{CH}_3(\text{CH}_2)_4\text{CH}=\text{CHCH}_2\text{CH}_2\text{CH}_2\text{COO}^-$

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Fatty Acids

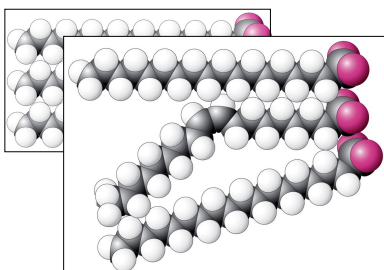
† Melting points are affected by chain length and the presence of cis double bonds.



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Fatty Acids

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Next up

+ Unit IV, Lecture 8 - Lipids and Cell Membranes,
cont'd. (Chapter 12)